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A computer simulation of mine air shaft thermodynamics

Ambyo Sumopandhi Mangunwidjojo

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A COMPUTER SIMULATION OF MINE

AIR SHAFT THERMODYNAMICS

by

AMBYO MANGUNWIDJOJO, 1935-

A DISSERTATION

Presented to the Faculty of the Graduate School of the

UNIVERSITY OF MISSOURI - ROLLA

In Partial Fulfillment of the Requirements for the Degree

DOCTOR OF PHILOSOPHY in. MINING ENGINEERING

Jughen Laugh Advisor

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ABSTRACT

Several earlier investigations have been conducted into various aspects of heat problems in underground mines. The application of general gas laws and thermodynamics in the approach of mine ventilation has been helpful in elucidating certain difficulties associated with the subject. By combining several of the techniques perfected by these researchers and applying carrier's method of calculation for heat flow from exposed rock strata into a workable plan, a method was developed whereby it is possible to accurately and conveniently predict the quality of working environment at any point within a deep mine.

The findings of this study may be of great value to designers in selecting economic optimums of labor, air conditioning and air transmission costs. They may be of further use in predicting underground conditions of fogginess, etc., which can be remedied before the mining operations begin.

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I. INTRODUCTION

The more lucrative, near surface ore deposits are gradually being exploited and consequently the mining of commercial minerals is extending deeper into the earth. As the operations penetrate greater depths, associated difficulties of economics, support and safety are amplified, sometimes to the verge of our technical capability. One of the most serious related problems confronting the mining engineer lies in predicting the atmospheric conditions that will be encountered, and particularly those that affect the comfort of miners. An extremely hot mine may be difficult or even impossible to operate from the standpoint of human endurance. Also, the natural tendency of heated air to flow without mechanical aid must be predetermined and integrated with the entire ventilation system in order to assure ^a positive flow control for normal operation and especially during emergencies.

The interdependent effects of time, depth, air autocompression, heat transferred by rock strata and mine water evaporation in conjunction with the seasonal and daily variations of intake air quality have yet to be evaluated in a satisfactory manner. The normally significant sources of mine heat have been identified and their respective contributions can be calculated for specific situations according to the principles of thermodynamics. Also, Carrier¹ has developed a method that relates the age of mine openings and the amount of heat flow from exposed rock strata. It is known, for example, that the rate of heat flow from exposed rock to air increases with depth and decreases with time. Also, it decreases with elevated intake air temperature. All of these phenomena are simultaneously influenced by the rate of mine water evaporation which is regulated by the quality of intake air, its resident time in the airway and the depth.

An accurate and useful determination or air properties at various key points in a mine ventilation circuit will involve the assignment of an initial intake air flow rate and quality, calculating the changes it will undergo in passing through an incremental length of airway of known conditions, adjusting its properties according to the thermodynamic changes invoked and repeating as before for new airway parameters of the next section. By such repetitive computations for all increments of an entire circuit, the air properties can be predicted for any point in the mine. Also, the entire procedure may be duplicated with appropriate adjustments to intake air quality to represent seasonal and daily changes and, furthermore, reduced strata heat flow, as a result of rock cooling, can be related to time in a similar manner. As a final goal, the thermodynamic properties of air can be ascertained for any point in the mine at any time during its operating life. However, because of the complicated, tedious and time-consuming procedure involved, this has never been accomplished for an actual mine to the best knowledge of this writer.

The purpose of this study is to alleviate the unfortunate situation described. As pointed out earlier, the pertinent thermodynamic principles are known and need only to be integrated into an accurate, workable method and the necessary computations reduced to a practical level. It is believed that both can be accomplished by appropriately assembling the known data and techniques into a suitable program for the)60/50 computer. The results are shown on the pages that follow.

II. LITERATURE REVIEW

A. Thermodynamics of Mine Air

1. Pertinent air properties

^Aclean atmosphere is a binary mixture of dry air and water vapor. On a volume basis, dry air contains 78.08 percent nitrogen, 20.95 percent oxygen and traces of approximately 15 other gases². The molecular weight of dry air is 28.966 (Appendix B1) and its gas law constant R_{a} is 53.34 ft-lbf/lbm-R. Water vapor has a molecular weight of 18.016 and its R_v value is 85.76 ft-lbf/lbm-R. Being a mixture, air does not behave as would a pure gas but, within the atmospheric limits of an operating mine, its departure from the gas laws is considered negligible. Water vapor normally exerts a low pressure and exhibits nearly perfect gas behavior.

Air flowing in a mine is never perfectly dry, it is always associated with water vapor. Due to evaporation, and often condensation, the weight of water vapor and consequently that of the dry air in a pound of atmosphere will change. It is therefore convenient when dealing with moist air to regard one pound of dry air as the fundamental unit and the flow is then considered as (14w) pounds of moist air, where w is specific humidity, the water vapor content in pounds associated with one pound of dry air.

The following equation of state expresses the relationship between pressure, volume and temperature:

$$
P \quad \mathbf{v} \quad = \quad R \quad T \tag{1}
$$

where
$$
P =
$$
 absolute pressure, $1bf/sq ft$.
\n $v =$ specific volume, cu ft/lbm.
\n $T =$ absolute temperature, ^oR.
\n $R =$ gas law constant, ft-lbf/lbm-R.

This expression is correct for every gas of a mixture each having its own values of P and R . Esbroeck³ suggested that the values of gas constants R are related to molecular weights M as

$$
R \mathbf{M} = 1545.4
$$

For mine air in coal mines composed of dry air (a}, water vapor (v), and methane gas or commonly known as firedamp $(d),$

$$
R_{a} M_{a} = R_{v} M_{v} = R_{d} M_{d} = 1545.4
$$
 (2)

and the R value for the mixture is related as

$$
G R_{m} = G_{a} R_{a} + G_{v} R_{v} + G_{d} R_{d}
$$

and

 $G = G_{a} + G_{v} + G_{d}$

where G is the total weight and G_{a} , G_{v} and G_{d} are partial weights of dry air, vapor and firedamp, respectively.

For specific humidity w, which is the weight of water vapor interspersed in each pound of dry air, the perfect gas equation may be applied as

lb of water vapor $\mathbf w$ lb of dry air

$$
w = \frac{p_v V / R_v T}{p_a V / R_a T} = \frac{p_v / R_v}{(P - p_v) R_a}
$$

$$
w = 0.622 \frac{P_V}{P - P_V}
$$
 (3)

from which

$$
p_V = \frac{W P}{0.622 + W}
$$
 (4)

where P is the barometric pressure and $p_{\mathbf{y}}$ and $p_{\mathbf{a}}$ are respectively the partial pressures of vapor and dry air at temperature T. Since

$$
P = p_a + p_v
$$

then

$$
P_{A} = P - \frac{w P}{0.622 + w} = \frac{P}{1 + 1.607 w}
$$
 (5)

2. Principles of psychrometry

A convenient relationship exists between vapor pressures, dry-bulb temperature, wet-bulb temperature and dewpoint temperature. The temperatures may be measured directly with appropriate instruments. Then, by reference to steam tables⁴. psychrometric tables or charts^{5.6.7}, all of the pertinent properties of moist air may be ascertained. Regnault⁸, in 1853, formulated an equation relating vapor pressure and dry- and wet-bulb temperatures in which

$$
p_{\mathbf{v}} = p_{\mathbf{s}, \mathbf{t} \mathbf{w}} - A P (t_{\mathbf{d}} - t_{\mathbf{w}})
$$
 (6)

where p_v = water vapor pressure, in.Hg.

$$
P_{S,tW} = \text{saturated water vapor pressure at } t_W, \text{ in.} \text{Hg.}
$$
\n
$$
P = \text{barometric pressure, in.} \text{Hg.}
$$
\n
$$
t_d = \text{dry-bulb temperature, } {}^{OF}.
$$
\n
$$
t_W = \text{wet-bulb temperature, } {}^{OF}.
$$
\n
$$
A = \text{pychrometric constant.}
$$

The value of A given by Regnault was 3.53×10^{-4} but has been modified by Barenburg⁸ to 3.613 X 10⁻⁴.

Air is usually deficient in water vapor and, therefore, its vapor pressure is lower than the maximum corresponding to the temperature of the mixture. The ratio of such vapor pressure to the vapor pressure of the mixture when saturated at the same temperature is termed the relative humidity ¢, that is

$$
\emptyset = 100 \frac{P_{\mathbf{V}}}{P_{\mathbf{S}}}
$$
 (7)

The relationship between specific humidity w and relative humidity \emptyset can be found by substituting the value of p_v of equation (7) into equation (3)

$$
w = \frac{0.622 \not Q \, p_S}{100 \, P - \not Q \, p_S}
$$
 (8)

This equation was used to plot various curves of relative humidity⁶.

At any temperature, air and its moisture has a certain enthalpy. Enthalpy of moist air H. in Btu/lbm of dry air, is the sum of the enthalpies of dry air H_{a} and water vapor H_v , or

$$
H = H_a + H_v
$$

and

$$
H = C_{pa} t_d + w (1061 + C_{pv} t_d)
$$
 (9)

Specific heat of dry air C_{pa} and of water vapor C_{pv} vary slightly with temperature as stated by Carrier⁵ in the expressions

$$
C_{pa} = 0.24112 + 0.000009 t
$$

and

$$
C_{\text{pv}} = 0.44230 + 0.00018 \text{ t}
$$

However, for all mining calculations the values of $C_{pa} = 0.24$ and $C_{pv} = 0.44$ Btu/lbm-^OR are sufficiently accurate and will therefore be used. By using the principle of mass fraction in a mixture, the specific heat of moist air can be found as follows:

$$
C_{\text{pm}} = \frac{C_{\text{pa}} + C_{\text{pv}} w}{1 + w}
$$
 (10a)

and

$$
C_{\text{vm}} = \frac{C_{\text{va}} + C_{\text{vv}} w}{1 + w}
$$
 (10b)

The thermodynamic aspect of airflow in mines will deal with the bulk of energy quantities. It involves the effects of changes of volume, temperature, pressure, humidity, and other parameters of the air as it flows underground. It is very useful in solving the problems of mechanical and natural ventilation in mines. The general energy equation for the flow of air of uniform composition and state between two points, from station 1 to station 2, may be stated as follows:

$$
J\left((H_1 - H_2) + Q_{1n} - Q_{out}\right) = \frac{U_2^2 - U_1^2}{2 g} + Z_2 - Z_1 - W_f
$$

$$
J\left((H_1 - H_2) + Q_{1n} - Q_{out}\right) = \int_2^1 \nabla dP - F_{1-2}
$$
 (11)

where H_1 and H_2 = enthalpies at stations 1 and 2, Btu/lbm. Q_{1n} and Q_{out} = heat added or lost by 1 lbm of air. U_1 and U_2 = air velocities, ft/sec. Z_1 and Z_2 = heights of stations above datum, ft. W_f = work done on the air by fans, ft-lbf/lbm. $J = Joule's$ equivalent of 778.26 ft-lbf/Btu. $v =$ specific volume, cuft/lbm. $P =$ absolute static pressure, lbf/sqft. F_{1-2} = work done against friction, ft-lbf/lbm. $g =$ acceleration due to gravity, ft/sec².

The above equation refers to the flow of one pound of dry air. For {1+w) pound of moist air, considering that the change of kinetic energy is negligible and with no fans in the airway, equation (11) is modified into

$$
(H_1 - H_2) + Q_{1n} - Q_{out} = \frac{(Z_2 - Z_1) (1 + w)}{J}
$$
 (12)

By considering the air flow process in the shafts as adiabatic due to auto-compression in downcast and expansion in upeast, then

$$
\int_{1}^{2} \mathbf{v} \ dP = J (H_2 - H_1) = (Z_1 - Z_2) (1 + \mathbf{w})
$$
 (13)

or

For the perfect gas relation,

$$
J (H2 - H1) = J Cpm (T2 - T1)
$$
 (14)

and thus

$$
J C_{\text{pm}} (T_2 - T_1) = (Z_1 - Z_2) (1 + w)
$$
 (15)

Equation (15) is used to find the final temperature of air flowing in a shaft of known depth, if the original state of air is known. For example, the temperature rise of air, having *0.00756* lb of water vapor per lb of dry air, for 1000 ft of depth will be:

$$
T_2 - T_1 = \frac{(Z_1 - Z_2) (1 + w)}{J \cdot C_{\text{pm}}} = \frac{1000 (1 + 0.00756)}{778.28 \times 0.2415}
$$

= 5.36 °F.

B. Sources of Mine Heat

The main sources of heat underground which will contribute to increasing mine air temperature are:

- 1. Auto-compression of the air.
- 2. Heat released by the rock strata.
-). Heat caused by blasting.
- 4. Heat caused by machinery, men, lighting and oxidation of rock, coal or timber.
- 5· Heat caused by friction of the air current.

The temperature increase caused by auto-compression and that issuing from strata are the major sources which contribute about 80 percent of all heat transferred to mine air⁹. Some is also contributed by rock movement¹⁰ although its relative amount is very incidental.

1. Temperature increase by air auto-compression

When air flows down a shaft into a mine, it is compressed at the rate of about 1 in.Hg per 900 ft of descent. The air is heated in the same way as if it were compressed in a compressor. If the shaft is dry, the temperature of the air will increase at the rate of about *5.5* op per 1000 ft of descent. If it is wet, the evaporation of moisture will modify the temperature-depth gradient as shown by equation (15). When no heat transfer takes place between the air and shaft, evaporation produces ^a fall in temperature.

According to the reversible adiabatic compression process,

 $p v$ ^k = constant

Since $P v = R T$ and considering stations 1 and 2 as shown by Fig. 1,

$$
\frac{P_1}{P_2} = \left(\frac{v_2}{v_1}\right)^{k} = \left(\frac{T_2}{T_1}\right)^{\frac{k}{1-k}}
$$

$$
P_2 = P_1 \left(\frac{T_2}{T_1}\right)^{\frac{k}{k-1}}
$$
 (16)

where k is the specific heat ratio $\frac{p_{\text{pm}}}{q}$, in which $\mathbf{v}_{\mathbf{v}\mathbf{m}}$ has been defined as specific heat of air at constant pressure and C_{vm} is the specific heat of air at constant volume. Values of C_{vm} may be obtained from the relation

$$
R = (C_{pm} - C_{vm}) J
$$

Equation (16) is used in the computer program to find the barometric pressure along the shaft.

Cooling due to evaporation is usually not sufficient to counteract the heating by compression. Thus, the air arrives at the downcast bottom with an increased temperature and absolute pressure and a reduced volume. The reverse processes occur in the upcast shaft where expansion takes place and work is done by the air to cause cooling. Hinsley¹¹ described the processes of the airflow in a mine circuit very similar to that occurring in a heat engine. The air is compressed in the downcast shaft, is heated in the levels and working areas where it expands and, in the upcast shaft, it expands still further due to the lowering of pressure. An indicator diagram shown in Fig. 1 represents the path of a pressure-volume change. The area enclosed by the P-v diagram indicates the quantity of work

Fig. 1 Indicator Diagram

in ft-lbf done during the cycle.

2. Heat released by the rock strata

The temperature of rocks forming the crust of the earth increases with depth. The base point for calculating the temperature rise is normally taken at about *50* ft below surface where the rock temperature remains at mean surface temperature throughout the year. The chief concern is to find the geothermic gradient, that is, the average increase in depth for a rise of 1 ^OF. Where the rock formation consists of several layers, the geothermic gradient will usually differ for each layer. Studies made in various parts of the world have shown that their values vary from one area to another and, also, with different depths^{12,13,14}. These variations are related¹⁴ to:

a. Mountains or folded rock structures.

b. Movement of water in the strata.

c. Local changes in radioactivity.

d. Erosion of the area.

e. Effect of glaciation in the past.

Geothermic gradient of rock can be explained by considering homogeneous and isotropic rock. Using one dimensional heat transfer, the conduction of heat through rocks follows the Fourier equation for steady conditions as follows:

$$
Q = - K_{r} A \frac{\Delta T}{\Delta Z}
$$

where $Q =$ rate of heat transfer by conduction, Btu/hr. $A = cross section normal to heat flow, sqft.$ ΔT = change in temperature, $^{\text{O}}\text{F}$. ΔZ = change in depth, ft. K_{r} = rock conductivity, Btu/hr-ft-^OF.

Assuming rock of constant thermal conductivity, the rate of heat transfer through a unit cross section is

$$
Q = \frac{K_{\mathbf{r}}}{Z} (T_{\mathbf{z}} - T_{\mathbf{0}})
$$

where T_z is the temperature of rock at depth z and T_o is the temperature of rock at the surface. Also,

$$
T_Z = (Q + \frac{K_T}{Z} T_O) \frac{Z}{K_T}
$$

and

$$
T_Z = T_O + \frac{Q Z}{K_T}
$$
 (17)

Thus, the temperature T_Z at depth Z depends on the surface rock temperature T_{0} , the depth, the rate of heat flow and the thermal conductivity of the rocks.

The factors which influence heat flow from the exposed rock strata of underground mine openings are specific heat, thermal conductivity, thermal diffusivity of the rock surrounding the mine openings and the heat transfer coefficient between the opening walls and the ventilating air. Specific heat C_r for most rocks is of the order of 0.2 Btu/lbm- $^{\text{OF}}$ ¹⁵. It varies slightly with temperature but, for this purpose, it may be regarded as being virtually constant for any given rock. Thermal diffusivity D, in sqft/hr, is related to the thermal conductivity and specific heat. Thus,

$$
D = \frac{K_r}{\int r^{-C_r}} \tag{18}
$$

where $\int_{\mathbf{r}}$ = rock density, lbm/cuft.

Heat from the high temperature walls is transferred into all mining openings. However, its rate of inflow decreases as a cooled, insulating layer of rock is built up around the opening. A study of several mines, conducted by Carrier¹, produced a chart of heat flow versus time which permits an accurate determination of the rate at which heat flows after various periods of cooling. The method has been developed mathematically for smooth, circular openings of infinite length. Carrier's chart shows a relationship between time and conductivity factors where the time factor is expressed by D t, in which D is the thermal diffusivity of rock and t is the time in years. Conductivity factor is given as the ratio of rock conductance and rock

conductivity, K/K_r where K represents the ability of rock to transmit heat expressed in Btu/sqft-hr-oF. To use the chart, the radius equivalent R_{e} of the airway must first be calculated,

$$
R_e = \frac{2 A}{P_m \sqrt{2}}
$$
 (19)

where $A = cross section of the shaft or airway, sqft.$ $Pm =$ perimeter of the airway, ft.

By a correlation-of-random-data method for curve-fitting, the general formula relating conductivity factor versus time factor as suggested by Carrier's chart was found to be

 $(n, 811)$ $n-2$ $A X$ $-(1.109)$ $n-2$ B) (20) $n-2$ -((1.109) $Y =$ where $Y =$ conductivity factor for radius of equivalent $R_e = n.$ $X = time factor = D t.$ $D =$ thermal diffusivity of rock, as calculated by equation (18) . $t = age$ of opening in years. $A = a constant = 0.0939034$ $B = a constant = 0.1956590$

Rock conductance K is found by multiplication of Y by

the rock conductivity $K_{\mathbf{r}}$.

$$
K = Y K_r \tag{21}
$$

If the rock surfaces are wet, a combination of convective and radiative heat transfer and evaporation take ^place. The degree of wetness can be specified in terms of wetness fraction f , which varies from zero for a perfectly dry surface to unity for thoroughly wet surface¹⁶. The combination of sensible heat flow for rock strata and the latent heat due to evaporation of water can be stated as

$$
J_{\mathbf{w}} = Pm L (K (t_{\mathbf{r}} - t_{\mathbf{a}}) + f r' E (p_{\mathbf{s},tr} - p_{\mathbf{v}}))
$$
 (22)

where
$$
J_{w}
$$
 = heat transmitted by rock surface, Btu/hr.
\n Im = perimeter of airway, ft.
\n L = length of airway, ft.
\nK = rock conductance, Btu/sqft-hr-^{OP}r.
\n t_{r} = rock surface temperature, ^{OF}r.
\n t_{a} = air temperature, ^{OF}r.
\nr' = latent heat of water evaporation, Btu/lbm.
\nE = coefficient of mass transfer, lbm/sqft-hr-1nHg.
\n $P_{s, tr}$ = saturated vapor pressure at t_{r} , in.Hg.
\n p_{v} = partial pressure of vapor, in.Hg.
\nf = degree of wetness of rock surface.

The value of E can be computed by using the formula¹⁷:

$$
E = \frac{h_c}{\int_A C_{pm} R_V T}
$$
 (23)

in which
$$
f_a
$$
 = density of mine air, lbm/cuft.
\n h_c = convective heat transfer coefficient,
\n $Btu/sqft-hr-OF$

The convective heat transfer coefficient h_c is calculated by using Mc Adams' formula¹⁷ which was based on experimental results for forced convection of turbulent flow. The formula was developed for smooth circular tubes:

$$
h_c = 0.023 \frac{K_a}{Dm} \left(\frac{G Dm}{u}\right)^{0.8} \left(\frac{u C_{pm}}{K_a}\right)^{0.4}
$$
 (24)

where K_{a} = thermal conductivity of air, Btu/ft-hr-^OF. $u =$ dynamic viscosity of air, lbm/ft-hr. C_{pm} = specific heat of air at constant pressure. $G =$ mass flow rate of air, lbm/sqft-hr. Dm = diameter of airway, ft.

Considering the size of mine air shaft diameter, it can be assumed that air shafts have smooth walls and thus, equation (24) can be applied. However, for underground mine airways, equation (24) has to be multiplied by a roughness factor F, which was determined by Starfield¹⁶ as 1.7. For airways with rectangular cross section, the value of Dm

$$
De = 4
$$

The values of K_{a} and u can be found in Appendix B and the thermal properties of some rocks are given in Appendix c.

3. Heat from other sources

a. Heat from blasting

The heat caused by blasting can be very great, about 300,000 Btu/hr10 or 3 Btu/hr per ton of monthly production in a metal mine. However, blasting is usually confined to ^asingle hour on one shift a day or to a certain period of each shift. This great amount of heat combined with the large amount of steam released by the explosives may adversely affect the ventilation system by impairing the existing heat balance of the mine atmosphere. The amount of heat and other products of blasting can be calculated by knowing the chemical composition of the explosives used and the amount consumed for each blast.

b. Heat from machinery

The amount of heat given to the air as a result of the use of power operated machines underground depends on the type of power used and on the kind of work done by each machine. The usual power systems employed in mines are compressed air and electricity. Exhausting compressed air is

usually cooler than the mine air thus it causes cooling. However, compressed air leaving a compressor at the surface is usually at an increased temperature compared with atmospheric air and when piped down a shaft there is usually a transfer of heat from the pipe line to the ventilating air. The heat lost by the compressed air in the shaft has been calculated from the approximate formula⁹:

$$
Q_{C} = G_{C} \left(0.24 \left(t_{1} - t_{2} \right) + 1061 \left(w_{1} - w_{2} \right) + \frac{(Z_{1} - Z_{2})}{778.26} \right) \tag{25}
$$

where $Q_c = \text{loss of heat, Btu/min.}$

- G_c = weight of compressed air flowing. lbm/min.
- t_1 = temperature of compressed air at top of shaft, ranging between 200 and 250 OF.
- t_2 = temperature of compressed air at bottom of shaft, ranging between *65* and 100 OF,
- w_1 = moisture content in the pipe at top of shaft.
- w_2 = moisture content in the pipe at bottom of shaft.
- Z_1 and Z_2 are the heights above some datum of the top and bottom of the shaft, respectively, ft.

For practical computations, the value of 1061 (w_1-w_2) is relatively small and, therefore, negligible.

For electrical machines used underground, the amount of heat given to the air can be computed on the basis of one horse power being equivalent to 42.72 Btu/hr¹⁰.

c. Heat from human metabolism

The chemical changes that take place in the human body produce heat. A miner working at his full normal capacity may produce heat at the average rate of 1100 Btu per hour¹⁰. Thus, assuming there are 300 miners underground working at an average rate, the heat output will be about JJO,OOO Btu/hr. Taking the volume of air flowing through the mine at JOO,OOO cuft/min., the temperature rise due to the human metabolism would be about 1 ^OF.

d. Heat from frictional losses

Frictional losses in the air stream are from two causes, those due to viscous drag in the thin laminar layer at solid boundaries and those due to turbulence. The heat caused by friction of air against the walls of the airways may be considered negligible. The air is heated by friction but the friction causes ^adrop in pressure and thus promotes an expansion of the air. The cooling by this expansion very nearly balances the heat produced by friction.

4. Evaporation of water

The dynamic equilibrium corresponding to the evaporation of water into air counterbalanced by the flow of heat from the air into the water is the basis of wet-bulb thermometry, a method used to determine the humidity of air. There is a saturated air film on the surface of water. This air film is at the same temperature as the water with which it is in contact. The transfer of heat and water vapor between the main air stream and the liquid water takes place through this saturated air film. Carrier⁵ stated that the increase in heat of vaporization is equa^l to the decrease in the sensible heat of the air stream from which the heat is extracted, that is,

latent heat absorbed = sensible heat lost

or

$$
r' (w' - w_0) = (C_{pa} + C_{pv} w_0) (t - t')
$$
 (26)

where
$$
r' =
$$
 latent heat of vaporization at the resultant
temperature t', Btu/lbm.

 w' = final weight of vapor per unit weight of dry air, lbm/lbm dry air.

$$
w_0
$$
 = initial weight of vapor per unit weight of
dry air, lbm/lbm dry air.

 $t' = final temperature.$

 $t =$ initial temperature.

$$
C_{pa}
$$
 and C_{pv} are specific heats of dry air and of the vapor respectively.

There is a continuous change in water vapor content of mine air along the ventilation circuit due to water evaporation. If there is no heat transfer between the air and the walls of mine openings, the evaporation produces a fall in temperature of the air. The temperature of equilibrium t' may be found by

$$
t' = t - \frac{r' (w' - w_0)}{C_{pa} + C_{pv} w_0}
$$
 (27)

as The rate of evaporation of water is given by Eckert¹⁸

$$
w_{\mathbf{v}} = \frac{h_{\mathbf{c}}}{C_{\mathbf{p}} \cdot R_{\mathbf{v}} \cdot T} (p_{\mathbf{s},td} - p_{\mathbf{v}})
$$
 (28)

where w_v = evaporation rate, lbm/sqft-hr.

$$
p_s, td
$$
 = saturated vapor pressure at t_d .
\n p_v = partial pressure of vapor, lbf/sqft.
\n R_v = gas constant for vapor, equal to 85.76.
\n T = temperature in Rankine.
\n h_c = heat transfer coefficient, equation (24).
\n C_p = specific heat of air per unit volume at
\nconstant pressure, equal to C_{pm} times the
\ndensity of air, Btu/cuft-OF.

c. Effects of Mine Heat

As mine air moves deeper into the bowels of the earth, it absorbs heat and moisture from various sources until its condition becomes humanly unbearable and the control of its flow difficult. Also the rock temperature increases at a rate of *0.5* to 3.3 °F per 100 ft of depth. Assuming a gradient of 1 °F per 100 ft, a mine that is

10,000 ft deep will suffer a rock temperature of 100 OF plus the mean annual surface temperature. This could be on the order of approximately 160 OF.

1. Effects upon miners

The productivity of an underground miner is related directly to his working environment. His body functions much the same as that of a machine. He takes on fuel in the form of food which his metabolism converts into energy for doing work. But unlike a machine, his system is only about 20 percent efficient. That is, for each unit of work performed, his body system must dispel about 4 equivalent units of heat. The rejected heat must be absorbed by the surrounding environment or, otherwise, the work rate will decrease (or cease) to prevent a heat buildup in the miner's body which eventually could cause prostration.

The approximate endurance capabilities of industrial workers are shown in Appendix A. A work efficiency versus comfort diagram is shown wherein comfort is represented by effective temperature, an interdependent combination of temperature, humidity and air velocity. An effective temperature chart depicting these relationships is also shown. To employ the diagrams, one needs to predetermine the wetand dry-bulb temperatures and air velocity. From these, the effective temperature may be obtained to ascertain the percent work efficiency that may be expected. Of course,

incoming air is being cooled and dehumidified mechanically to promote a more productive working environment. However, it is not presently possible to accurately predict the amount of conditioning required. Instead, air conditioners of several thousand tons capacity are selected on the basis of empirical rules. Very often, extensive later modifications entail considerable additional expense.

Because the air conditioning demands will change daily, seasonally and with time as the mine rock cools, an accurate prediction of air quality during successive stages of its operating life would be of great value. Such data would show the cooling load extremes and a practical mean by which a single unit eould be selected.

2. Effects upon natural ventilation

Most ventilation systems in underground mines are ^a combination of mechanically and naturally induced flow whether or not the natural aspect was ever considered in the initial design. A few air conditioned mines possess natural air pressures of such magnitude as to entirely eliminate the need for mechanical prime movers. Should the natural flow phenomenon be overlooked in an air circuit layout, future control of the system could prove difficult. For example, the writer has repeatedly observed ^a complete reversal in flow direction after a routine blast. Toxic fumes were routed into the mine instead of to the upcast shaft. Even more serious occurrences have been

recorded during fires and other emergencies.

Natural ventilation is produced essentially by ^a thermodynamic process in which heat is converted into work. Thermal energy added to the system is converted into a pressure head capable of producing airflow. Warm humid air rises to displace cooler, drier air. Contaminating, low density gases will enhance the effect. The actual flow tendency within a mine is dependent upon the relative conditions of incoming and outgoing air. Also, a sudden influx of cold or hot air may disrupt a delicately pressure balanced air circuit on a specific mining level.

One of the methods for calculating the natural ventilation pressure h_n is by density. This involves the consideration of two stationary vertical columns of air, which constitute the downcast and upcast as in Fig. 2, and treating the system as a standard manometer. Where the surface levels of the shaft collars are not the same, the upper common level of the columns will be the level of the higher shaft collar. The lower level of the two columns is the deepest point reached by the air current.

Assuming that the air is at rest so that the full weight of air in each shaft is acting on its respective base, B is the absolute atmospheric pressure in lbf/sqft at the upper level joining the tops of the air columns and L the height of columns in ft., the natural ventilation

of Natural Ventilation
pressure of the system can be calculated by

$$
h_{n} = (B + L W_{d}) - (B + L W_{u})
$$

= L (W_{d} - W_{u}) (29)

where h_n = natural ventilation pressure, lbf/sqft. W_d = mean density of air in downcast shaft, and W_{11} = mean density of air in upcast shaft, lbm/cuft.

As can be appreciated, the mean air densities are difficult to predict. However, it is known that the density of air along the shaft changes according to the following?:

$$
W = \frac{1.3258 (P - 0.3777 P_v)}{t_d + 459.69}
$$
 (30)

where P and p_v are expressed in in.Hg. By basing the calculations on incremental sections along the shaft, the air densities of each section of the shaft can be computed. Then by averaging the densities in each column, the values of W_d and W_u can be found.

III. DISCUSSION OF PROBLEM

Several investigations^{1,7,8,16,19} have been conducted into various phases of the problem and each has contributed in some degree toward formulating the procedures that follow. By accepting standardized techniques of these earlier researchers, combining them with some of the gas laws and thermodynamics into a workable plan and then modifying the whole for computer programming, the writer was able to attain the goals outlined earlier.

A. Method of Computation

In attacking the problem, it was first assumed that: 1. The rock is homogeneous and isotropic.

- 2. The shaft is circular in cross section (can be modified) •
- .3. The air temperature is uniform in each cross section,
- 4. Intake air quality does not fluctuate during a computation.
- *5.* The shaft walls are wet and underground airways have ^a wetness fraction of 0.6 (can be modified).
- 6. The effects of enthalpy change in each section increment of shaft and airways are instantaneous.
- 7. Mine air behaves as a perfect gas. The use of the existing formulas discussed in the literature review for the computer simulation was based on this assumption,
- 8. Air flows in mine ventilation circuit as ^aturbulent flow.

Consideration of the mine air circulation system was then divided into four convenient parts as shown in Fig. 3: 1. Definition of air quality on the surface.

2. Enthalpy change in the downcast shaft.

3. Enthalpy gain on the mining level.

4. Enthalpy change in the upcast shaft, with special note of the point at which condensation commences.

It was assumed that 200 miners were working on a mining level as schematically outlined in Fig. 4.

At the outset of the program, all necessary basic data were read into the computer. This included tables of saturated vapor pressures and latent heats of water vaporization, shaft dimensions, rate of air flow, rock properties, conditions and dimensions of the mine workings, age of openings, surface air conditions, etc. Having assigned wet- and dry-bulb temperatures and an appropriate barometric pressure to the surface intake air, all of the remaining pertinent thermodynamic properties of the atmosphere were computed by reference to input data and methods previously described in the review of literature.

The surface air, with its condition defined, was then assumed to pass the initial 5-foot increment of the shaft. Its enthalpy changes were computed according to the parameters previously assigned in input data. These changes involved the effects of:

1. Auto- compression.

2. Water evaporation.

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Fig. 4 Schematic of an Underground Working Plan

 $\bar{\mathcal{A}}$

3. Heat from the exposed strata.

4. Compressed air lines.

Having the heat and water exchange determined, the air condition was redefined accordingly and passed through the next 5-foot segment of shaft where the process is duplicated for new shaft parameters in accord with the geothermic gradient. By repetitive computation, the entire depth of air column is treated in this manner to provide complete thermodynamic data at its base or at any desired increment along its length. At specified intervals the computer will print out the air conditions along the shaft.

The air is next moved through the working level where its enthalpy changes involve:

1, Heat from exposed rock.

2. Water evaporation.

3. Compressed air lines, machinery, workers, lights, etc. Compressed air, in the working areas, has a cooling effect. For any complicated underground network, the amount of air flow through each airway can be found by using the Hardy-Cross method of ventilation network analysis²⁰. Knowing the specific heat of the air, total sensible heat transferred to the air per unit time and the amount of air passing in the same time interval, the dry-bulb gradient of the air can be calculated. A new condition is assigned to the air at the base of the upcast shaft according to the heat and water exchanges undergone.

In the upcast shaft, the procedure duplicates that

used in downcast but with proper consideration for a negative geothermic gradient and auto-expansion. Also, it was assumed that compressed air lines were not present. In addition to the usual data, the point at which condensation commences was ascertained. This is useful in combatting shaft water problems. The exhaust mine air in upcast shaft usually has higher temperatures compared to the surrounding walls, therefore, it can be expected that the age of opening has little effect on the amount of heat transmitted by rock strata.

Incremental densities of air in each of the shafts were averaged, converted to column base pressures and subtracted to produce a measure of the natural ventilation pressure existing in the mine. Fig. *5* shows the flow chart of the program and Appendices D-1, D-2 and D-J list the computer input and output data.

B. Results

The results of this study are gratifying. As anticipated, air quality can be accurately and rapidly ascertained at any key point within a mine and natural ventilation pressures can be determined simultaneously as shown by the appended computer print-out and Tables 1, 2, J and 4.

Knowing the depth of mine, required air velocity, rock properties and range of surface air quality, the underground environmental conditions can readily be predicted. Also, because of the ease and rapidity of calculation, any

J6

Fig. *5* Computer Flow Chart

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number of determinations may be made to represent the situation that will exist during any period during the life of the mining operation. This should simplify the tedious task of air conditioner selection and flow system design,

Several computations were conducted to test the environmental response to various input air conditions, The resulting data compared very favorably to solutions derived mathematically and graphically for similar conditions, In addition, the mine atmospheric parameters varied in accordance with the laws of thermodynamics throughout the air course. Some of these displayed significant trends which may justify further discussion.

The rise in dry-bulb temperature with increasing depth is practically linear as shown in Tables 1 and ² and Fig, 6. It would appear that because of increasing rock temperature with depth, the dry-bulb gradient would also increase. However, in examining equation (22), it can be seen that there is an increase in enthalpy gain as suspected but there is also an increase in sensible heat conversion to latent heat in evaporating more water. This is further substantiated by Table 4 wherein the more saturated incoming air displays a larger (very slightly) dry-bulb gradient.

Higher incoming air and rock temperatures appear to

)9

TABLE 1, EFFECTS OF AIR VELOCITY ON

AIR CONDITION ALONG DOWNCAST SHAFT *)

1990年 - 1990年

Dry-bulb = 50 ^OF Wet-bulb = 48 ^OF

*) shaft diameter: 24 ft.

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TABLE 2. EFFECTS OF AIR VELOCITY ON

AIR CONDITION ALONG DOWNCAST SHAFT *)

Dry-bulb = 75 ^{OF} Wet-bulb = 60 ^OF

*) shaft diameter: 24 ft.

Fig. 6 Temperature Gradients vs Depth

have no practical effect upon the rate or increase.

Wet-bulb temperatures, when plotted against depth (Fig. 6), display a gradient that is minutely upwar^d curving, However, the change in its depression below the dry-bulb will increase with depth for more saturated incoming air and decrease with drier intake. This in accord with equations (28) and (6) which show the rate of evaporation and the wet-bulb depression both being dependent upon the existing degree of saturation. Table 4 and Fig. 7 demonstrate the effect of relative humidity of air input on the dry-bulb and wet-bulb gradients. By varying the wet-bulb temperatures while maintaining the dry-bulb as constant, the effect of relative humidity of the incoming air on the dry-bulb and wet-bulb gradients can be seen. Dry air shows a steep wet-bulb gradient due to more water pick-up as suggested by equations (6) and (28) and thus reducing the wet-bulb depression significantly.

Natural vetilation pressure, N.V.P., results from a difference in air densities between the downcast and upcast shafts. As would be expected; cold, dry intake air will invariably result in a high pressure difference between the two columns. Hot, dry air entering the mine may result in a higher N.V.P. than cool, humid atmosphere since humidity reduces density in much the same manner as does increasing temperature. This is demonstrated in Table 3 and according to equation (30).

TABLE 3. EFFECT OF TEMPERATURE VARIATION OF

AIR INPUT ON NATURAL VENTILATION PRESSURE *)

**) Age of working areas is 0.1 year

4. EFFECTS OF RELATIVE HUMIDITY TABLE ON TEMPERATURE GRADIENTS *)

Drybulb Wetbulb Rel.Hum. Depth Drybulb Wetbulb Rel.Hum. OF OF Х OF oF X ft 55.000 40.000 19.63 50.000 70.77 55.000 Surface 67.54 60.402 54.322 $60, 398$ 47.134 33.47 1000 65.763 58.784 65.57 65.759 53.599 42.74 2000 63.365 64.41 59.612 49.01 71.127 71.121 3000 76.494 68.043 63.77 76.487 65.319 53.33 4000 63.47 70.812 56.35 81.865 72.791 81.858 5000 77,602 63.38 87.240 58.51 87.234 76.163 6000 68.000 75.000 70.23 53.000 19.09 75,000 Surface 72.342 67.96 80.407 33.47 61.860 1000 80.397 76.856 66.58 69.446 43.17 85.777 85.764 2000 81.504 65.77 49.79 91.152 76,228 3000 91.137 86.258 65.32 82.476 96.534 4000 96.517 54.35 101.926 91.095 65.12 88.369 57.55 101,911 5000 65.06 95.996 59.81 94.025 107.325 6000 107.310

*) Geothermic gradient of rock strata: 0.75 OF/100 ft descent Age of airway: 1.0 year, surface rock temperature: 65ºF.

Fig, 7 Effects of Relative Humidities on Wet-bulb Gradients

 $\frac{4}{5}$

As air passes through a mine shaft at an increased velocity, the rates of heat exchange and evaporation are increased. However, the enthalpy and wet-bulb gradients are lowered as shown in equation (9) and Fig. 8 respectively. That is, the amounts of heat and water absorbed per unit time are greater but the effects upon each poun^d of dry air decreased. This is because the larger air volume is less affected by the shaft conditions and, as ^a result, is more amenable to heat and water absorbtion. These behavior are in accord with equation (28).

Relative humidity, in its response to depth, is dependent to a major extent upon the conditions of input air. Cold, saturated intake atmosphere will cause a decreasing relative humidity as the air flows deeper. The reverse holds true for hot, dry air entering the mine. This situation is the same as described earlier for wetbulb temperature. The two are interdependent and should therefore react in a similar manner.

In reviewing the above phenomena associated with the original purpose of this study, it appears that the designer of air conditioning and flow systems for deep mines may utilize these findings to grea^tadvantage. For example, they can be useful in seeking economic optimums between N.V.P. (which reduces the required mechanically induced pressure) and the air conditioning load (which affects N.V.P. and the rate of strata cooling) while main-

 \bar{z}

Fig. 8 Effects of Air Velocity and Depth on Wet-bulb Gradient

 $8[†]$

 \bullet

taining a suitable environment in the working areas.

The ideal cooling load must also be considered with respect to the age of the air course, Initially the strata will be hot and the mine air will require considerable conditioning whereas in time the rock will be cooled and the load reduced, An air conditioner selected to fit the original demands will be too large in the final stages. A smaller unit will cause an initial decrease in work efficiency but also a reduction in air treatment costs. This, coupled with the potential advantages of N.V.P., poses ^aproblem in economics that has plagued the design engineer for decades.

Of lesser importance but still significant is the shaft size. Heretofore, this has been fixed at an optimum between installation and air transmission costs. Now, because of the ease in predicting its influence upon the rates of enthalpy gain and strata cooling, the shaft size is destined to play a more important role in mine system design. Other, indirect effects of environmental control, such as fogginess, spalling, etc., may also become involved but their predictions can be accomplished without difficulty.

^Acomputer simulation of the many mine environmental situations that are possible for a given ore deposit can be utilized in predicting the various related expenses that will prevail for each. It seems, therefore, that it

is now practical to balance the per ton (of ore mined) costs of labor, air conditioning and air circulation to an overall minimum.

The point where condensation commences in the upcas^t shaft will depend on intake air conditions and the conditions in the underground levels. The point was found by the same method of heat exchange computations used in the downcast shaft. It can be expected that the age factor of the upcast shaft does not have any effect in the computations due to the fact that the exhaust mine air has higher temperatures compared to the surrounding walls. This situation has been incorporated in the computer program. Fig. 10 shows the effect of air velocity on the point of saturation of air along the upcast shaft.

Fig. 9 Effect of Air Velocity on Moisture Content

IV. CONCLUSIONS AND RECOMMENDATIONS

A. Conclusions

The procedures followed and the results obtained during the course of this study have already been discussed. However, a review of the findings indicates that the following conclusions may be drawn:

- 1. A computer simulation method has been developed whereby the quality of underground environment and the related N.V.P. may be accurately and conveniently predicted for any predefined combination of mine circumstances.
- 2. The effects of air input rate and quality and shaft size upon N.V.P. and air conditioning load may be determined by the same technique.
- J. The above procedures can be of great assistance to designers in optimizing the per ton costs of labor, air conditioning and air circulation.
- 4. Underground conditions conducive to foginess, rock spalling, reverse air flow, etc., can be predicted and remedied during the early stages of design.
- B. Recommendations

The method described has been developed in theory only. It should therefore be tested against existing mine

conditions before its practicality can be accepted. The writer has spent some months in a deep mine with this very purpose in mind but due to circumstances beyond his control. the project had to be abandoned. Initiation of an alternate experiment was prevented by lack of time.

Should this investigation be continued, it will soon become apparent that the computer simulation program has been developed to permit simple modification to fit conditions that may exist in nearly any mine. However, the computer outputs will be no more accurate than the data supplied.

APPENDIX \mathbf{A}

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Effective Temperature Scale and the Effect of Effective Temperature on Work Efficiency

APPENDIX B

1. Composition of Dry Air

Source: Reference (2)

2. Property Values of Dry Air

t = temperature $V =$ kinematic viscosity Pr = Prandtl \hat{r} = density k = conductivity number $C_p =$ specific heat D = diffusifity

APPENDIX C

1. Geothermic Gradient for

Different Mining Districts

Source: Reference (8)

2. Conductivity and Density of Rocks

Source: Clark, Jr., s.P. Handbook of Physical Constants

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APPENDIX D

- 1. Input Data and Symbols
	- for the Computer Program

 $SUBOCK$ = annual mean temperature of rock at surface, in $^{\textcircled{\tiny G}}$ F.

GRAD1, GRAD2 and GRAD3= geothermic gradient of strata. CONDR1, CONDR2 etc. $=$ rock thermal conductivity. ROCDS1, ROCDS2 etc. = density of rock. CPROCK $=$ specific heat of rock. $TABLE(I)$. ETABLE(I) = table of latent heat of evaporation. $TABLE(I)$, $BTABLE(I) = table of saturated vapor pressure.$ KKEY, KEY $=$ symbols for repetition of program. CPAIR and CPVAP $=$ specific heat of air and vapor.

APPENDIX D

2. Output of Computer Program^{*})

APPENDIX D-2 (Cont.)

and the company

AIR CONDITIONS AT JUNCTION 1 DRYBULB= 107.37 WETBULB= 95.01 AIR CONDITIONS AT JUNCTION 4 DRYBULB= 107.66 WETBULB= 95.01
BOTTOM OF UPCAST SHAFT, DRYBULB= 110.60 WETBULB= 97.51

APPENDIX D-2 (Cont.)

AVERAGE DENSITY OF UPCAST AIR = 0.077379

NATURAL VENTILATION PRESSURE = 1.821151 INCHES W.G.

```
*) 
  The input data used for the computations: 
  Shaft diameter: 24 ft. 
  Air flow input: 500000 cuft/minute. 
  Geothermic gradient of strata: 1.5 °F/100 ft descent.
  Surface rock temperature: 65 OF 
  Input air conditions: drybulb = 75 <sup>OF</sup>, wetbulb = 61 <sup>O</sup>F.
                          barometric pressure at surface = 29.92 in.Hg.
  Age of shafts: 1,0 year, age of stope: 0.2 year. 
  The actual output data were printed out for every 50 ft depth interval in downcast 
  shaft and 20 ft depth interval in the upcast shaft.
```
APPENDIX D-2 (Cont.)

Symbols used in Appendix D-2, Computer Output:

- $A =$ Depth, $ft.$
- ^B= Barometric Pressure, lbf/sqin.
- $C =$,, ,, , in.Hg.
- $D = Dry-bulb$ Temperature, $^{\text{OF}}$.
- $E =$ Wet-bulb Temperature, $^{\text{O}}F$.
- $F =$ Wet-bulb Depression, ${}^{O}F$.
- G = Wet-bulb Gradient, ${}^{0}F/100$ ft.
- $H =$ Specific Volume, cuft/lbm.
- ^I= Density, lbm/cuft.
- $J =$ Moisture Content, lbm vapor/lbm dry air.
- $K =$ Enthalpy, Btu/lbm.
- $L =$ Relative Humidity, per cent.

APPENDIX D

3. Computer Program

```
C *COMPUTATIONS IN DOWNCAST SHAFT* 
        COMMON TABLE(130),ETABLE(130),BTABLE(1JO) 
        DIMENSION TWET(600) 
      KKEY=0<br>READ(1,210)DIAM,AIBCFM,DEPTH,STRAT1,STRAT2<br>READ(1,205)LAYERS<br>READ(1,220)SUROCK,GRAD1,GRAD2,GRAD3,CONDR1,CONDR2,CONDR3,ROCDS1,<br>1ROCDS2,ROCDS3,CPROCK
        READ(1,230)AGE1,AGE2<br>READ(1,199)(TABLE(I),ETABLE(I),I=1,119)<br>READ(1,299)(TABLE(I),BTABLE(I),I=1,119)<br>READ(1,200)DTEMSF,WTEMSF,SAPDCS,SAPUCS,SURPIN
  199 FORMAT(F5,0,F7.2) 
  299 FORMAT(F5.0,F9.4) 
   200 FORMAT(5F10.3) 
   205 FORMAT(I3) 
   210 FORMAT(5F10.2) 
   220 FORMAT(7F6.),JF6,2,F6.3) 
   230 FORMAT(2F10.3) 
   800 WRITE(3.215)DTEMSF, WTEMSF, SAPDCS, SAPUCS
   215 FORMAT(//10X, 'ATMOSPHERIC CONDITIONS: DRYBULB=', F6.2, ' F, ', 1X,
      1'WETBULB=',F6.2,' F',/10X,'BAR.PRESSURE: DOWNCAST=',F7.3,'PSIA.',
       21X, 'UPCAST=', F7, 3, 'PSIA'')<br>WRITE(3,225)DIAM,AIRCFM
   225 FORMAT(10X, 'DIAMETER OF SHAFT=', F5.2, ' FEET.', 2X, 'AIR INPUT=',
       3F9.1, ' CUFT/MIN'/)
        WRITE(3,224)LAYERS 
   224 FORMAT(10X, 'NUMBER OF LAYERS=', I3/)<br>IF(LAYERS-1)172, 172, 173
   172 GRAD2=GRAD1 0
```
GRADJ=GRAD1 GO TO 174 173 IF(LAYERS-2)172,175,174 175 GRAD3=GRAD2 174 WRITE(3,235)SUROCK, GRAD1, GRAD2, GRAD3 235 FORMAT(10X, 'SURFACE ROCK TEMP.=', F6.2, ' F, ', 1X, 'GEOTHERMIC GRADIEN 4T LAYER1=',F4,2,' F/100 FT DESCENT'/59X,'LAYER2=',F4,2,' F/100 FT 5DESCENT'/59X,'LAYER3=',F4.2,' F/100 FT DESCENT'/) WRITE(3,245)AGE1.AGE2 245 FORMAT(10X,'AGE OF AIRWAY=',F6.3,' YEARS,',1X,'AGE OF STOPE=',F6.3
6.' YEARS'/) IF(LAYERS-1) 72,72,73 72 STRAT1=DEPTH GO TO 76 ⁷³IF(LAYERS-2)72,74,75 74 STRAT2=DEPTH 76 ST2aSTRAT2-STRAT1 $IF(ST2-0.0)57,57,58$ $57 \text{ } ST2 = 0.0$ $ST3=0.0$ GO TO 150 58 ST2=ST2 $ST3=0.0$ GO TO 150 7 *5* STRAT3=DEPTH ST3=STRAT3-STRAT2 ST2=STRAT2-STRAT1 150 WRITE(),226)DEPTH,STRAT1,ST2,STJ 226 FORMAT(10X, 'MAXIMUM DEPTH OF SHAFTS=', F8.2, ' FEET, ', 1X, 'THICKNESS ?OF LAYER1=',F8.2,' FEET'/62X, 'LAYER2=',F8.2, I FEET'/62X, 'LAYER3=', 8F8.2,' FEET'////) $WRTTE(3,100)$ ¹⁰⁰FORMAT(//50X,'DOWNCAST SHAFT'//) WRITE(3,101) 101 FORMAT(6X,'DEPTH',?X,'PRESSURE',5X,'TEMPERATURE',2X, 'WETBULB',1X, *0*

```
1'WETBULB', 1X, 'SPEC. VOL. ', 1X, 'DENSITY', 2X, 'MOIST. CONT.',
     1'ENTHALPY', 1X, 'REL.HUM.')
      WRITE(3.102)
 102 FORMAT(44X, 'DEPRESS', 1X, 'GRADIENT', 19X, '(LB/LB OF')
      WRITE(3,103)
  103 FORMAT(7X, '(FT)', 4X, '(PSIA)', 2X, '(INHG)', 1X, '(DRY F)',
     2'(WET F)', 2X, '(F)', 3X, '(F/100FT)', '(CUFT/LB)', '(LB/CUFT)',
     31X, 'DRY AIR) ', 2X, '(BTU/LB) ', 3X, '(%) '//)
      *CONSTANTS USED IN CALCULATIONS
\mathbf CCPVAP = 0.44CPAIR=0.24RCAIR = 53.345RCVAP = 85.760RCABTU=0.06854RCVBTU=0.11019
      CVVAP=CPVAP-RCVBTU
      CVAIR=CPAIR-RCABTU
      AIRCON=0.0155AIRVIS=0.0435\mathbf{C}DTEMSR=DTEMSF+459.69
       WTEMSR=WTEMSF+459.69
      T = DTEMSFCALL TABEL (T, VAPOR)
       SATPS1 = VAPORT=WTEMSF
       CALL TABEL (T.VAPOR)
       SATPSW=VAPOR
       PVAP1=SATPSW-0.0003613*SURPIN*(DTEMSF-WTEMSF)
       PV1PSI = PVAP1*0.491W=(0.622*PV1PSI)/(SAPDCS-PV1PSI)DEN=1.3225*(SURPIN-(0.3777*PVAP1))/DTEMSR
       SPVOL=1. / DEMCPM = (CPAIR+W*CPVAP)/(1.+W)CVM = (CVAIR + W * CVVAP) / (1. + W)
```
 \mathfrak{L}
$CK=CPM/CVM$ RELHUM=100. * PVAP1/SATPS1 ENTHAL=CPAIR*DTEMSF+W*(1061.+CPVAP*DTEMSF) DEP-DTEMSF-WTEMSF WRITE(3,104)SAPDCS, SURPIN, DTEMSF, WTEMSF, DEP, SPVOL, DEN, W, ENTHAL, 1 RELHUM 104 FORMAT (6X, 'SURFACE', 1X, F7. 3, 2X, F5. 2, 1X, F7. 3, 1X, F7. 3, 1X, F6. 3, 2X, $2'0.000'$, 2X, F8.3, 3X, F8.6, 2X, F9.7, F8.2, 2X, F6.2) $SECTS=5.00$ AREA=3.14*DIAM*SECTS $CROS=(3.14*DIAM**2.)/4.$ AIRATE=AIRCFM/CROS VOLUM=CROS*SECTS FLOW=SECTS/(AIRATE*60.) $PERIM=3.14*DIAM$ $DELTAZ=0.$ $DLTBTU=0$. CUMBTU=0. TOTDEN=0. $SUMDEN = 0.$ $X = 1000$. $L=1$ $N=1$ TDUMMY=WTEMSF+0.001 *CALCULATIONS OF CORRECTION FOR AGE OF OPENINGS* \mathbf{C} CONSTA= 0.0939034 $CONSTB=0.195659$ RADIUS=2. * CROS/(1.414*PERIM) A=RADIUS-2. $IF(A-0.0)8.9.9$ $8 \text{ A} = A^*(-1)$ $S=1./(0.811**A)$ $Q=1./(1.109**A)$ GO TO 7 $9 S=0.811***$ A $Q=1.109$ **A

 ∞

CALCULATIONS OF HEAT TRANSMITTED BY COMPRESSOR LINE \mathbf{C} 7 COMPRS=2000. $TCOMP1=250.$ $TCOMP2=100.$ BTUCOM=COMPRS*DEN*(0.24*(TCOMP1-TCOMP2)+DEPTH778.26) \mathbf{C} 25 DO 5 I=1.10 DELTAZ=DELTAZ+SECTS TEMP=DELTAZ*(1.4W)/(CPM*778.26)+DTEMSF TEMPR=TEMP+459.69 $P =$ SAPDCS*(TEMPR/DTEMSR)**(CK/CK-1.)) $PIN=PI*29.92/14.696$ $AIBDEN=144.*(P-PV1PSI)/(RCAIR*TEMPR)$ DST=1.3225*(PIN-(0.3777*PVAP1))/TEMPR TOTDEN=TOTDEN+DST AIRFLO=(CROS*AIRATE*60.*DST)/CROS CP=CPM*DST CONVEC=(0.023*AIRCON/DIAM)*((AIRFLO*DIAM/AIRVIS)**0.8)*(AIRVIS*CPM $4/AIRCON$) ** 0.4 BETA=CONVEC/CP $T = T EMP$ CALL TABEL (T.VAPOR) $SATPS1 = VAPOR$ DELTAW=BETA*70.7259*(SATPS1-PVAP1)/(RCVAP*TEMPR) DWATER=DELTAW*AREA*FLOW WDRAIR=CROS*60.*AIRATE*FLOW*AIRDEN $WATER=$ ($W*WDRAIR$) + $DWATER$) / $WDRAIR$ $W = WATER$ $CPM = (CPAIR+WATER*CPVAP)/(1.+WATER)$ CVM=(CVAIR+WATER*CVVAP)/(1.+WATER) $CKM = CPM/CVM$ TRMASS=CONVEC/(DST*CPM*RCVAP*TEMPR) PV2PSI=WATER*P/(0.622+WATER) PV1PSI=PV2PSI PVAP2=PV2PSI*29.92/14.696

 $PVAP1 = PVAP2$

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- TFKTOR=AGE1*DIFFUS CFKTOR=(S*CONSTA)/(TFKTOR**(CONSTB*Q)) GO TO 143 3 IF(DELTAZ-50.)49,49,50 49 ROCKTF=SUROCK GO TO 43 50 IF(DELTAZ-STRAT1)51,51,52 51 ROCKTF=SUROCK+(GRAD1/100.)*(DELTAZ-50.) $ROCKT1 = ROCKTF$ GO TO 43 52 IF(DELTAZ-STRAT2)53,53,54 53 ROCKTF=ROCKT1+(GRAD2/100.)*(DELTAZ-STRAT1) ROCKT2=ROCKTF GO TO 46
- 46 CONDUC=CONDR2 ROCDSTeROCDS2 DIFFUS=CONDUC/(ROCDST*CPROCK)
-
- $ROCKT1 = ROCKTF$ GO TO 43 48 ROCKTF=ROCKT1+(GRAD2/100,)*(DELTAZ-STRAT1)
- 45 IF(DELTAZ-STRAT1)47,47,48 47 ROCKTF=SUROCK+(GRAD1/100,)*(DELTAZ-50,)
- 44 ROCKTF=SUROCK GO TO 43
- 43 CONDUC=CONDR1 ROCDST=ROCDS1 DIFFUS=CONDUC/(ROCDST*CPROCK) TFKTOR=AGE1 *DIFFUS CFKTOR=(S*CONSTA)/(TFKTOR**(CONSTB*Q)) GO TO 143 2 IF(DELTAZ-50.) $44, 44, 45$
-
- GO TO 43 42 ROCKTFaSUROCK+(GRAD1/100.)*(DELTAZ-50.)
- 41 ROCKTF=SUROCK
- GO TO (1,2,3),LAYERS 1 IF(DELTAZ-50.)41,41,42

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- 54 ROCKTF=R0CKT2+(GRAD3/100.)*(DELTAZ-STRAT2) CONDUC=CONDR3 ROCDST=ROCDS3 DIFFUS=CONDUC/(ROCDST*CPROCK)
TFKTOR=AGE1*DIFFUS $CFKTOR=(S*CONSTA)/(TFKTOR**(CONSTE*)$) 143 T=ROCKTF
- CALL TABEL(T, VAPOR)
SATROC=VAPOR WALL=PERIM*SECTS*(CFKTOR*CONDUC*(ROCKTF-TEMP)+(ELATEN*TRMASS* 3(SATROC-PVAP1)))
DBTU=WALL*FLOW DLTBTU=DLTBTU+DBTU
HEAT FROM COMPRESSOR LINE
- C*HEAT FROM COMPRESSOR LINE*
BTUCPR=BTUCOM*FLOW*60,
DLTEM=BTUCPR/(WDRAIR+DWATER)*CPM TEMP=TEMP+DLTEM

CALL TIBLE(T,ELATEN)

- C **HEAT FROM EXPOSED ROCK STRATA** IF(ROCKTF-TEMP)93,93,95
	- 93 DTEMP=(DWATER*ELATEN)/(WDRAIR+DWATER)*CPM TNET1=TEMP-DTEMP
		- TEMP=TNET1
		- GO TO 84
		- *95* DLTT=DBTU/{WDRAIR+DWATER)*CPM TNET2=TEMP+DLTT TEMP=TNET2
		- 84 T=TDUMMY CALL TABEL(T, VAPOR)
XSAT=VAPOR VALUE=XSAT-0.0003613*PIN*(TEMP-TDUMMY)
IF(PVAP1-VALUE)86.86.85
		- 85 **TDUMMY=TDUMMY+0.00115**
			- GO TO 84
		- 86 GO TO *5*
		- *5* CONTINUE

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 $TWET(N)=TDUMMY$ **TF=TEMP** $SKOL=1. / DST$ RLHUM=100. *PVAP1/SATPS1 ENTLPY=CPAIR*TEMP+WATER*(1061.+CPVAP*TEMP) CUMBTU=CUMBTU+DLTBTU AVEDEN=TOTDEN/10. SUMDEN=SUMDEN+AVEDEN TOTDEN=0. DLTBTU=0. IF(DELTAZ-X)11, 37, 11 11 WRITE(3,105)DELTAZ, P, PIN, TF, TWET(N), SEVOL, DST, WATER, ENTLPY, RLHUM 105 FORMAT (5X, F8. 2, 1X, F7. 3, 2X, F5. 2, 1X, F7. 3, 1X, F7. 3, 16X, F8. 3, 23X, F8.6, 2X, F9.7, F8.2, 2X, F6.2) $N = N + 1$ IF(DELTAZ-DEPTH) 25, 35, 35 $37 J=20*(L-1)$ $QQ=0.$ $V = TWET(N)$ DO $38 \text{ I} = 1.19$ $K = I + J$ $N=K$ $U = TWET(N+1) - TWET(N)$ 38 QQ=QQ+U $RAD=2. *QQ/19.$ DPS=TEMP-V $L = L + 1$ $X = X + 1000$. $N = N + 2$ WRITE(3,106)DELTAZ, P, PIN, TF, V, DPS, RAD, SEVOL, DST, WATER, ENTLPY, RLHUM 106 FORMAT(5X, F8.2, 1X, F7.3, 2X, F5.2, 1X, F7.3, 1X, F7.3, 1X, F6.3, 2X, F5.3, 3X, 6F7.3,3X,F8.6,2X,F9.7,F8.2,2X,F6.2) IF(DELTAZ-DEPTH)25,35,35 35 AVEDST=(SUMDEN+DEN)/121. WRITE(3,109)AVEDST 109 FORMAT(5X, 'AVERAGE DENSITY OF DOWNCAST AIR=', F8.6)

COMPUTATIONS IN UNDERGROUND LEVELS \mathbf{C} $JNOTON=1$ READ(1,410) AERCFM, CROS1, PRIME, LENGTH 410 FORMAT (3F10.2.15) $LSECT = 10$ WTEMP-TDUMMY $DLTBTU=0.$ $CUMBYU=0.$ TTIME=0. $KEY = 0$ DIFFUS=CONDUC/(ROCDST*CPROK) TFKTOR=AGE2*DIFFUS REDIUS=2. * CROS1/(1.414*PRIME) 450 DIAMT=4. *CROS1/PRIME AERATE=AERCFM/CROS1 **SECT=LSECT** AFLOW=SECT/(AERATE*60.) AAREA=PRIME*SECT*0.6 AA=REDIUS-2. IF(AA-0.0)308,309,309 308 AA=AA*(-1.) $SS=1./(0.811***AA)$ $SQ=1./(1.109**AA)$ GO TO 310 309 SS=0.811**AA SQ=1.109**AA 310 CFKTOR=(SS*CONSTA)/(TFKTOR**(CONSTB*SQ)) **T=ROCKTF** CALL TIBLE(T.ELATEN) CALL TABEL (T.VAPOR) SATROC=VAPOR TDUMMY=WTEMP+0.001 N=LENGTH/LSECT $\mathbf C$

DO 55 I=1.N $P = P$

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PIN = PINTEMPR=TEMP+459.69
   CPM = (CPAIR+WATER+CPVAP)/(1.+WATER)CVM = (CVAIR+WATER*CVVAP)/(1.+WATER)AERFLO=(CROS1*AERATE*DST)/CROS1
   CP \leftarrow CPM \ast DSTCONVEC=1, 7*(0.023*AIRCON/DIAMT)*((AERFLO*DIAMT/AIRVIS)**0.8)*
  6(AIRVIS*CPM/AIRCON)**0.4
   BETA=CONVEC/CP
   T = T EMPCALL TABEL (T.VAPOR)
   SATPS1 = VAPORT=WTEMP
   CALL TABEL (T.VAPOR)
   SATPSWEVAPOR
   PVAP1=SATPSW-0.0003613*PIN*(TEMP-WTEMP)
   DELTAW=BETA*70.7259*(SATPS1-PVAP1)/(RCVAP*TEMPR)
   DWATER=DELTAW*AAREA*AFLOW
   PV1PSI = PVAP1 * 0.491AERDEN=144. * (P-PV1PSI)/(RCAIR*TEMPR)
   WDRAER=CROS1*60. * AERATE* AFLOW* AERDEN
   WATER=((WATER*WDRAER) + DWATER) / WDRAER
   DST1 = 1.3225*(PIN-(0.3777*PVAP1))/TEMPRTRMASS=CONVEC/(DST*CPM*RCVAP*TEMPR)
   WALL=PRIME*SECT*(CFKTOR*CONDUC*(ROCKTF-TEMP)+(ELATEN*TRMASS*
   5(SATROC-PVAPI))DST = DST1DBTU=WALL*AFLOW
    DLTBTU=DLTBTU+DBTU
   IF(ROCKTF-TEMP) 593.593.595
593 DTEMP=(DWATER*ELATEN)/(WDRAER+DWATER)*CPM
    TNET1=TEMP-DTEMP
    TEMP=TNET1
    GO TO 584
595 DLTT=DBTU/(WDRAER+DWATER)*CPM
    TNET2=TEMP+DLTT
```
XSAT=VAPOR

VALUE=XSAT-0.0003613*PIN*(TEMP-TDUMMY)

IF(PVAP1-VALUE)586,586,585

585 TDUMMY=TDUMMY+0.00115 GO TO 584 *586* GO TO *55 55* CONTINUE DISTAN=LENGTH TIME=DISTAN/AEBATE BTU=DLTBTU/TIME CUMBTU=CUMBTU+DLTBTU TTIMEaTTIME+TIME WTEMP=TDUMMY DLTBTU=O. C *COMPUTATIONS OF HEAT PROBLEMS FOR DIFFERENT UNDERGROUND AIRWAYS* IF(KEY-1)458,457,457
458 TTERMN*TEMP WRITE(3,851)JNCTON,TTERMN,WTEMP GO TO 456 457 JNCTON=JNCTON+1
WRITE(3,851)JNCTON,TEMP,WTEMP GO TO 456 851 FORMAT(/5X,'AIR CONDITIONS AT JUNCTION', I3, 1X, 'DRYBULB=', F7.3.' F' $2, 2X, 'WETBULB=' , F7.3,' F')$ 456 KEY=KEY+1
IF(KEY.EQ.1) GO TO 460 IF(KEY.EQ.2) GO TO 470
IF(KEY.EQ.3) GO TO 480
IF(KEY.EQ.4) GO TO 490
IF(KEY.EQ.5) GO TO 495
IF(KEY.EQ.6) GO TO 496
IF(KEY.GT.6) GO TO 497 460 READ(1, 410) AERCFM, CROS1, PRIME, LENGTH

TEMP=TNET2

CALL TABEL(T,VAPOR)

584 T=TDUMMY

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GO TO 450

- 470 READ(1,410)AERCFM, CROS1, PRIME, LENGTH GO TO 450
- 480 READ(1,410)AERCFM,CROS1,PRIME,LENGTH TSTOPE=TEMP MSTOPE=THMASS
	- PSTOPE=SATROC
	- VSTOPEaPVAP1 GO TO 450
- 490 READ(1, 410) AERCFM, CROS1, PRIME, LENGTH GO TO 450
- 495 READ(1, 410) AERCFM, CROS1, PRIME, LENGTH GO TO 450
- 496 READ(1,410)AERCFM, CROS1, PRIME, LENGTH GO TO 450
- 497 AIR)=AERATE DST3=DST CPM]=CPM TEMWET=TDUMMY TEMDRY=TEMP
	- DPRS=TEMDRY-TEMWET
- C *ADDITIONAL HEAT SOURCES*
- C *200 MINERS WORKING UNDERGROUND, 18.4 BTU/MIN PER MINER* CURENT=20000. WORKER=200.*18.4 ALLAMP=100.
	- COMPRS=8000.
- C *STOPE: 30 X 100 FT, THREE WORKING AREAS*
	- $CROS=30. *10.$ PERIM=80.
	- $AIRCUM=50000.$
	-
	- SECT=100.
AIRATE=AIRCUM/CROS
	- $FLOW = SECT/(AIRATE*60.)$
	- WALL= PERIM*SECT*(CFKTOR*CONDR3*(ROCKTF-TSTOPE)+(ELATEN*MSTOPE* \downarrow
4(PSTOPE-VSTOPE)))
	-

WALL=3.*WALL DBTU=WALL*FLOW CUMBTU=4.*CUMBTU HOKBTU=CUMBTU/TTIME TTLBTU•CURENT+WOHKEH+ALLAMP-COMPHS+ROKBTU AIR23=AIR3*12,*12.*DST3 BTUAIR=TTLBTU/AIR23 $ADTEMP = BTUAIR/CPM3$ DRYST3=TTERMN+ADTEMP DRYGRD=(DPRS/TEMDRY)*DRYST3
WETST3=DRYST3-DRYGRD WRITE(3,501)DRYST3, WETST3 501 FORMAT(//5X,'BOTTOM OF UPCAST SHAFT, DRYBULB=',F7.3,2X,'WETBULB=', $2F7.3$)
T=DRYST3 CALL TABEL (T, VAPOR) SATSIN=VAPOR WRITE(3.600) 600 FORMAT($\frac{7}{750X}$, 'UPCAST SHAFT' $\frac{7}{7}$) 601 FORMAT(6X,'DEPTH',7X,'PRESSURE',5X,'TEMPERATURE',2X,'DENSITY',2X, 6'MOIST,CONT. '} WRITE(J,602) 602 FORMAT(7X,'(FT)',8X,'(INHG)',8X,'(DRY,F)',4X,'(LB/CUFT)',1X,

7'(LB/LB,DRY AIR)'//)

DRST3R=DRYST3+459.69

DRYST4=(DRYST3-(DEPTH/(CPM*778.26)))

DRST4R=DRYST4+459.69

PRES3=SAPUCS*(DRST3R/DRST4R)**(CKM/(CKM-1.)) PRES3=SAPUCS*(DRST3R/DRST4R)**(CKM/(CKM-1.))
PRIN3=PRES3*29.92/14.696 T=WETST3 CALL TABEL(T, VAPOR) SATPV3=VAPOR
PVST3=SATPV3-0.0003613*PRIN3*DRYGRD
PV3PSI=PVST3*0.491 PVST3=SATPV3-0,0003013*Fhin3*DhionD
PV3PSI=PVST3*0,491
DNS3=1,3225*(PRIN3-(0,3777*PVST3))/DRST3R

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WSTA3=(0,622*PV3PSI)/(PRES3-PV3PSI)<br>RELAT=100,*PVST3/SATSIN<br>ENTAL3=CPAIR*DRYST3+WSTA3*(1061.+CPVAP*DRYST3)<br>WRITE(3,610)DELTAZ,PRIN3,DRYST3,DNS3,WSTA3
   610 FORMAT(5X,F8,2,6X,F5.2,9X,F7.3,4X,F8,6,5X,F9.7) 
C *COMPUTATIONS IN UPCAST SHAFT* 
            Z34=0, TOTDNS=0.
            SUMDNS=0.
            ROCKUP = ROCKTF900 DO 615 I=1, 4Z34=Z34+SECTS<br>TMPUP=DRYST3-(Z34*(1.+WSTA3)/(CPM*778.26))<br>TMPUPR=TMPUP+459.69
            PRSUP=PRES3*(TMPUPR/DRST3R)**(CKM/(CKM-1.))<br>PRSUPI=PRSUP*29.92/14.696<br>AIRDST=144.*(PRSUP-PV3PSI)/(RCAIR*TMPUPR)<br>DNSUP=1.3225*(PRSUPI-(0.3777*PVST3))/TMPUPR<br>TOTDNS=TOTDNS+DNSUP<br>AIRFLO=(CROS*AIRATE*60.*DNSUP)/CROS
           CPU=CPM*AIRDST<br>CONVEC=(0.023*AIRCON/DIAM)*((AIRFLO*DIAM/AIRVIS)**0.8)*(AIRVIS*<br>2CPM/AIRCON)**0.4<br>BETAUP=CONVEC/CPU
             T=TMPUP
             CALL TABEL(T,VAPOR) SATUP<del>=</del>VAPOR<br>DLTWAT=BETAUP*70.7259*(SATUP-PVST3)/(RCVAP*TMPUPR)<br>ADWAT=DLTWAT*AREA*FLOW
             WDRYR=CROS*60.*AIRATE*FLOW*AIRDST<br>WATRUP=((WSTA3*WDRYR)+ADWAT)/WDRYR<br>WSTA3=WATRUP
             CPM<del>=</del>(CPAIR+WATRUP*CPVAP)/(1.+WATRUP)<br>CVM=(CVAIR+WATRUP*CVVAP)/(1.+WATRUP)<br>CKM=CPM/CVM<br>TRMASS=CONVEC/(DNSUP*CPM*RCVAP*TMPUPR)
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c

PV4PSI=WATRUP*PRSUP/(0,622+WATRUP) PVJPSI=PV4PSI PVAP4=PV4PSI*29.92/14.696 PVST3=PVAP4 ZUP=DEPTH-Z 34
IF(LAYERS.EQ.1) GO TO 631 IF(LAYERS.EQ.2) GO TO 632
IF(LAYERS.EQ.3) GO TO 633 631 ROCUP=ROCKUP-(GRAD1/100.)*Z34
638 CONDUC=CONDR1 ROCDST=ROCDS1
DIFFUS=CONDUC/(ROCDST*CPROCK)
TFKTOR=AGE1*DIFFUS CFKTOR=(S*CONSTA)/(TFKTOR**(CONSTB*Q)) GO TO 651 632 IF(STRAT1-ZUP)641,6J9,649 641 ROCUP=ROCKUP-(GRAD2/100.)*234 658 CONDUC=CONDR2 ROCDST=ROCDS2
DIFFUS=CONDUC/(ROCDST*CPROCK) TFKTOR=AGE1*DIFFUS CFKTOR=(S*CONSTA)/(TFKTOR**(CONSTB*Q)) GO TO 651 639 ROCUP1=ROCUP GO TO 649 649 ROCUP=ROCUP1-(GRAD1/100.)*(STRAT1-ZUP) GO TO 638 6JJ IF(STRAT2-ZUP)644,646,645 644 ROCUP=ROCKUP-(GRAD3/100.)*Z34 CONDUC=CONDR3 ROCDST=ROCDS3
DIFFUS=CONDUC/(ROCDST*CPROCK)
TFKTOR=AGE1*DIFFUS CFKTOR=(S*CONSTA)/(TFKTOR**(CONSTB*Q)) GO TO 651 646 ROCUP2=ROCUP

GO TO 676 676 ROCUP=ROCUP2-(GRAD2/100.)*(STRAT2-ZUP) GO TO 658 645 IF(STRAT1-ZUP)646,6J9,649 651 T=ROCUP CALL TIBLE(T,ELATEN) CALL TABEL (T.VAPOR)
SATRUC=VAPOR WALLUP=PERIM*SECTS*(CFKTOR*CONDUC*(ROCUP-TMPUP)+(ELATEN*TRMASS* 3(SATRUC-PVST3)))
DBTU=WALLUP*FLOW
DLTBTU=DLTBTU+DBTU IF(ROCUP-TMPUP)693,693,695 693 DTEMP=(ADWAT*ELATEN)/(WDRYR+ADWAT)*CPM
TNETU1=TMPUP-DTEMP TMPUP=TNETU1 GO TO 615 695 DLTT=DBTU/(WDRYR+ADWAT)*CPM
TNETU2=TMPUP+DLTT TMPUP=TNETU2 T=TMPUP
CALL TABEL(T.VAPOR) IF(PVST3-VAPOR) 615, 612, 612
612 PV3PSI=PVST3*0.491 $W\$ STA3=(0.622*PV3PSI)/(PRSUP-PV3PSI)
615 CONTINUE AVEDNSTOTDNS/4.
SUMDNS=SUMDNS+AVEDNS TOTDNS=0. T=TMPUP
CALL TABEL(T, VAPOR) IF(PVST3-VAPOR) 667,668,668 ⁶⁶⁷WRITE(3,671)ZUP,PRSUPI,TMPUP,DNSUP,WSTAJ 671 FORMAT(5X,F8.2,6X,F5.2,9X,F7.3,4X,F8.6,5X,F9.7)
IF(Z34-DEPTH)900,666,666 ⁶⁶⁸WRITE(J,672)ZUP,PRSUPI,TMPUP,DNSUP,WSTAJ,PVSTJ

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672 FORMAT(5X,F8.2,6X,F5.2,9X,F7.3,4X,F8.6,5X,F9.7,2X,'*SATURATED*',<br>
12X,'VAP.PRES.=',F9.4)<br>
IF(Z34-DEPTH)900,666,666<br>
666 AVDSTY=(SUMDNS+DNS3)/301.
     VENAT=(Z34/5.2)*(AVEDST-AVDSTY)<br>WRITE(3,661)AVDSTY
661 FORMAT(5X,'AVERAGE DENSITY OF UPCAST AIR=',F8.6)<br>WRITE(3,662)VENAT
662 FORMAT(5X, 'NATURAL VENTILATION PRESSURE=', F8.6, ' INCHES W.G.')
     KKEY=KKEY+1 
     IF(KKEY.EQ.1) GO TO 992 
     IF(KKEY.EQ.2) GO TO 993
     IF(KKEY,GT,2} GO TO 997 
992 READ(1,200)DTEMSF, WTEMSF, SAPDCS, SAPUCS, SURPIN
     GO TO 800 
993 READ(1,200)DTEMSF, WTEMSF, SAPDCS, SAPUCS, SURPIN
      GO TO 800 
997 STOP 
      END 
      SUBROUTINE TIBLE(T,ELATEN) 
      COMMON TABLE(130),ETABLE(130),BTABLE(130) 
      J=0IT = TDO 17 I=1,119 
      J=J+1<br>IF(TABLE(I)-IT)17,18,17
  17 CONTINUE 
  18 ELATEN=ETABLE(J) 
      RETURN
       ————————————————————
      END 
      SUBROUTINE TABEL(T,VAPOR) 
      COMMON TABLE(130),ETABLE(130),BTABLE(130) 
       J=0
```
DO 27 $I=1,119$ $J=J+1$ IF(TABLE(I)-T)27,28,29

- 27 CONTINUE
- 28 VAPOR=BTABLE(J)
	- GO TO 40
- 29 RATIO=(BTABLE(J)-BTABLE(J-1))/(TABLE(I)-TABLE(I-1))*(TABLE(I)-T) VAPOR=BTABLE(J)-RATIO
- 40 RETURN

END

VI. BIBLIOGRAPHY

- 1. Carrier, W.H. "Air Cooling in the Gold Mines on the Rand," American Inst. Mining Engineers-Technical Publication, n. 970, September 1938.
- 2, Threlkeld, J.L. Thermal Environmental Engineering. New Jersey: Prentice-Hall, Inc., 1962.
- 3. Esbroeck, Guy van. "Thermodynamics of Mine Ventilation," Colliery Engineering, Part 1, February 1950, 68-72, Part 2, April 1950, 149-153, Part 3, September 1950, 366-369.
- 4. Keenan, J.H., and Keyes, F.G. Thermodynamic Properties of Steam. New York: Wiley & Sons, Inc., 1964.
- 5. Carrier, W.H. "Rational Psychrometric Formulae," Transactions, A.S.M.E., v. 33, 1911, 1005-1027.
- 6. Goodman. W. Air Conditioning Analysis. New York: The MacMillan Company, 1943.
- 7. McElroy, G.E. Bureau of Mines Report of Investigation, R.I. 4165, December 1947.
- 8. Parczewski, K.I., and Hinsley, F.B. "Hygrometry in nuites," Transactions, Institution of Mining Engineers.
v. 116, 1956-1957, 64-81.
- 9. Oakes, A.A., and Hinsley, F.B. "Heat and Humidity in a Deep Coal Mine," Transactions, Institution of Mining Engineers, v. 115, 1955-1956, 52-78.
- 10. Forbes, J.J., Davenport, S.J., and Morgis. G.G. Review of Literature on Conditioning Air for Advancement of Health and Safety in Mines, Bureau of Mines, I.C. 7528 (1949) .
- 11. Hinsley, F.B. "A New Method of Evaluating the Effects of Natural Agencies on the Ventilation of Mines," Transactions, Institution of Mining Engineers, v. 97- 98, 1938-1939. 131-151.
- 12. Biswas, N. "Study of Heat and Humidity in Mines, Part ¹ Thermocouple Psychrometry in Mines," Colliery Engineering, August 1964, 337-344.
- 13, Mullins, R., and Hinsley, F.B, "Measurement of Geothermic Gradients in Boreholes," Transactions, Institution of Mining Engineers, v. 117, 1957-1958, 380-393.
- 14, Bromilow, J,G, "Conditioning of the Ventilating Air in Coal Mines," Transactions, Institution of Mining Engineers, v. 116, 1956-1957, 538-557.
- Scott, D.R. "The Cooling of Underground Galleries," $15.$ Transactions, Institution of Mining Engineers, v. 118, 1959. 355-373.
- 16. Starfield, A.M., and Dickson, A.J. "A Study of Heat Transfer and Moisture Pick-up in Mine Airways," Journal of the South African Institute of Mining and Metallurgy, December 1967, 211-234.
- 17. Rohsenow, W.M., and Choi, H.Y. Heat, Mass and Momentum. New Jersey: Prentice-Hall, Inc., 1963.
- 18. Eckert, E.R.G. Introduction to the Transfer of Heat and Mass. New York: McGraw-Hill Book Company, Inc., 1950.
- 19. Sharp, D.F. "Sources of Moisture in Mine Airflow," International Journal of Rock Mechanics and Mining Sciences, v. 4, 1967, 71-83.
- 20. McPherson, M.J., "Mine Ventilation Network Problems-Solution by Digital Computer," Colliery Guardian, August 21, 1964, 253-259.

VII. VITA

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