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MECHANICAL AND NEUTRONIC PROPERTIES OF NEUTRON ABSORBING MATERIALS

BY

DONALD JAMES BUTH, 1963-

A THESIS

Presented to the Faculty of the Graduate School of the

UNIVERSITY OF MISSOURl-ROLLA

In Partial Fulfillment of the Requirements for the Degree

MASTER OF SCIENCE IN NUCLEAR ENGINEERING

1987

Approved by

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PUBLICATION THESIS OPTION

This thesis has been prepared in the style utilized by Nuclear Technology. Pages 1-43 will be presented for publication in that journal. Appendices A - E have been added for purposes normal to thesis writing.

ABSTRACT

The primary neutron absorbing material examined in this project was BoraflexTM, which consists of a silicone polymer with boron carbide acting as the neutron absorber. Two aluminum based neutron absorbers, Boral $^{\texttt{TM}}$ and Al-B-Ten, were also examined. Both X-ray and neutron radiography were used to examine the BoraflexTM. The neutron radiography gave better resolution in detecting fluctuations in the boron carbide distribution; however, X-ray radiography did perform better in detecting flaws in the base material. Tensile tests were performed to determine the uniformity of BoraflexTM's tensile properties. The average tensile strength was calculated to be $303 + 15$ psi. A small angular dependence was found in the material; however, the tensile properties were uniform as a whole. Neutron activation analysis was used to identify the presence of trace impurities in the Boraflex TM that could become radioactive after long exposures to neutrons. The elements identified were magnesium, manganese, zinc, dysprosium, and lanthanum. The total removal cross section for thermal neutrons, and the absorption cross sections were also determined. The total cross section for BoraflexTM was 14 \pm 1 barn. This was compared to BoralTM, which had a cross section of 220 \pm 20 barn. The absorption cross section for the Boraflex T_M was $12 + 1$ barn. This value was compared to the absorption cross section of Al-B-Ten, which was 4.8 ± 0.5 barn.

ACKNOWLEDGMENTS

The author would like to thank all of the staff members at the University of Missouri-Rolla Reactor Facility for their assistance in conducting the neutron attenuation and trace element analysis experiments, and to Don Alger and Dave Peeler for their help with the neutron radiographs performed at the Missouri University Research Reactor.

Special thanks are extended to my committee members, Dr. Albert Bolon, Dr. D. R. Edwards, and Prof. R. V. Wolf for their help in conducting these experiments.

The author is grateful for the BoraflexTM and Al-B- $\texttt{Ten}^{\texttt{TM}}$ materials that were supplied by BISCO Industrial Services, Inc. He is also grateful for the financial support he received as a United States Department of Interior Fellow and as a Power Fellow.

And many thanks to my wife, Lisa, who has given me support and encouragement throughout this research.

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I. INTRODUCTION

This research involved the examination of various properties of a material called Boraflex™, and the comparison of some of these properties, when possible, to other neutron absorbing materials. Boraflex TM is a composite shielding material for neutrons that is currently being used in high density fuel storage racks. It is made by adding boron carbide (B_4C) powder to a silicone polymer [1]. After the polymer solidifies it is cut into sheets and then clad in stainless steel to form the storage racks.

Boron carbide is used as a neutron absorbing material for many reasons. It contains approximately 80 atomic percent natural boron, which has two isotopes 10 B (18.8%) and 11_B (81.2%). The 10_B has a very high neutron absorption cross section (3836 barns) for thermal neutrons. Because of this, the B_4C also has a large cross section for thermal neutrons. Other properties that make B_4C very attractive are its hardness (Knoop Hardness Number 27.5 GPa), high melting point (2350 $^{\circ}$ C), and chemical inertness. However, B_4C performs poorly under loading because it has a relatively low tensile strength, and is very brittle [2].

Since the 1940's this problem has been avoided by combining the B_4C , in a powdered state, with other more ductile materials [3] to give a composite material that has better mechanical characteristics, while still retaining the neutronic properties of the B_4C . Aluminum, steel, polymers,

and various resins have all been used successfully for this purpose [1-7]. This paper reports the results of a study to determine the mechanical and neutronic properties of a polymer based neutron absorber, and compares the neutronic properties with some aluminum based neutron absorbers. More information on high density fuel storage racks and some of the alternate materials discussed in this report are presented in articles by Weeks [8], Burn [9], Macmillan [10], Evans [11], Judd [12], and the Electric Power Research Institute [13].

II. EXPERIMENTAL METHODS AND MATERIALS

A. OVERVIEW

The principle material that was examined in this research was BoraflexTM. It is produced by BISCO Products, Inc. for use in high density fuel storage racks, and in other neutron shielding applications. The main characteristics that were examined were the B_4C distribution, the uniformity of the tensile properties, the impurity content, and the absorption and total thermal neutron cross sections. The experimental cross sections were also compared to the total cross section for Boral TM , and the absorption cross section of a new absorbing material, called Al-B-Ten TM , that consists of B_4C in aluminum.

B. TEST MATERIALS

Three materials were examined in this report. The primary material was BoraflexTM. It is a composite material consisting of powdered B_4C in a methylated polysiloxane elastomer matrix. The material has a total thickness of 0.078 inches, and a boron loading of 0.020 $q/cm²$ 10 B.

Boral TM , a product of Brooks and Perkins Advanced Structures, was used as a reference material because of its widespread use as a neutron absorbing material. It is made by mixing powdered B_4C with approximately 50% of the required aluminum powder and then combining this mixture with molten aluminum. This mixture is then cast as ingots and sprayed with aluminum to ensure that no B_4C is exposed. A 0.125 inch sheet of 2S aluminum is then wrapped around the material for cladding, and it is rolled, while still warm, to give a final thickness of either 0.25 or 0.125 inches $[5,6]$. The BoralTM used for these experiments was 0.25 inches thick with a cladding thickness of 0.020 inches [6], and a 10 B loading of 0.101 g/cm².

The third material that was examined was $A1-B-Ten^{TM}$. It is a mixture of 40 volume percent B_4C in aluminum which is then sandwiched between two layers of pure aluminum that are approximately 5 to 10 mils thick. It has a total 10 B loading of approximately 0.03 g/cm^2 and a plate thickness of 0.10 inches [14].

C. RADIOGRAPHY

Both neutron and X-ray radiography were used to determine the B_4C distribution in the polymer. Both of these techniques use a source of radiation to pass through an object and then strike a film, which produces a negative image of the original object on the film. For the neutron radiography either a direct conversion or a transfer method is normally used. In the direct conversion method the neutrons pass through the object and then strike a conversion screen. This screen gives off low energy X-rays, which then expose the film. For the transfer method, the neutrons pass through the object and activate a metallic plate. This radioactive plate is then placed in contact with a piece of

film for a certain length of time in order to expose the film. This method is especially useful for radiographing radioactive materials. For X-ray radiography, the X-rays directly expose the film so that no conversion screen or transfer plate is required. However, intensifying screens are sometimes used to either give a better resolution at the same exposure, or to limit the exposure needed to give a certain film density. Each of these methods have some definite advantages and disadvantages over the other one; however, a best technique does not always exist. In many cases both techniques can be used to supplement the information from the other.

1. Neutron Radiography The first technique examined, neutron radiography, can be very useful in differentiating between two elements with similar atomic numbers, as long as the materials have a difference in their absorption cross sections. Also, neutron radiography can be used to identify flaws in a material that is encased in a metal shroud. This characteristic allows materials to be examined even after they have been clad in steel or lead. The disadvantages for the neutron radiography are that the process can be very expensive, especially if long exposure times are required, the material examined may become radioactive for a period of time, and there are not many neutron radiographic facilities available [15,16,17].

The neutron radiographs were made using the neutron

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radiographic facility located on the thermal column of the Missouri University Research Reactor (MURR) at Columbia, Missouri. The beam characteristics for these radiographs include a thermal flux of $6.2*10^7$ n/cm²s, a collimator length to diameter ratio of 38.83, and a gamma dose rate of 1.2 mR/s [14]. A 12.5 micron gadolinium conversion screen and M5 Kodak Industriex film was used to produce the radiographs through direct thermal neutron radiography. Samples 2, 6, 10, and 14 were attached to the front of a loaded 356mm by 432mm vacuum film cassette holder, which was then inserted into the sample chamber. The film was exposed to the neutrons for 1.5 hours before being removed to be processed using chemicals and procedures shown in reference 15. The finished radiographs were analyzed using a Macbeth TD-504 densitometer, and a kodak color densitometer.

2. X-ray Radiography The other method used to examine the BoraflexTM was X-ray radiography. X-Ray radiography has been used for non-destructive examination of many different materials since the early 1900's [17]. Because of this, there are many places available which can perform X-ray radiography at relatively low costs. The main disadvantages with X-ray radiography are its poor ability to distinguish between elements with very close atomic numbers, and its inability to radiograph objects that are radioactive.

The X-ray radiographs were made using a Faxitron X-ray machine with a tube to object distance of 25.5 inches.

Samples 2, 6, 10, and 14 were used, after they had been checked for any induced radioactivity, so that any artifacts found by either examination method could be compared with the other technique. A tube voltage of 20 kilovolts was used with an exposure of 1.25 mA-minutes. Dupont NDT 75 radiographic film was used with a 0.01 inch lead intensifying screen. The radiographs were developed using a 5-10-15 developing technique using Kodak GBX chemicals. This technique consist of a 5 minute immersion in developer followed by a 10 minute immersion in a fixer solution and ending with a 15 minute water wash. Because the samples were larger than the available film, each sample had to be radiographed in halves. This required two radiographs to be made for each sample, but by using external reference marks, the location of any artifact could be identified. The radiographs were analyzed using a Kodak Model 1 color densitometer.

D. TENSILE PROPERTIES

Tensile tests were performed on BoraflexTM samples in both pre- and post-irradiated conditions. These tests were designed to determine the uniformity of the tensile properties of the BoraflexTM, and to determine if these properties were isotropic. Samples were also tested to see if a small stress raiser would greatly affect the tensile properties. For these tests ASTM E-21 was followed as closely as possible, with the major difference being that

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the size of the tensile specimen used was smaller than the one recommended. This procedure determined the engineering stress and percent elongation by using short term static loading. The engineering stress is defined by the equation

$$
Stress = F / A_0 \qquad (Eq. 1.)
$$

and the percent elongation is given by

$$
\$ \text{ Elongation} = (1 - 1_0) / 1_0 \qquad (\text{Eq. 2.})
$$

is the applied load where F

- is the original cross sectional area \mathbf{A}_{\odot}
- is the final gage length 1

is the original gage length 1_{Ω}

The stress raiser was a small cut in the neck of the sample. This cut was perpendicular to the length of the tensile specimen and reduced the width of the neck by 0.1 cm. It was utilized because of the difficulty involved in detecting cuts in the BoraflexTM with either of the radiographic techniques, or through visual inspection. This induced stress raiser is not in accordance with the ASTM procedure being used; however, it should not adversely affect the validity of the test results since stress raisers mainly influence fatigue tests [19].

To determine if the BoraflexTM had an angular dependence in its tensile properties, it was decided to perform tensile tests on specimens that were at 0^0 , 45^0 , 90^0 , and 135⁰ from the normal direction of the BoraflexTM sheet. Straps 2, 3, 5, 6, 7, 9, 11, 13, and 15 were used for the unirradiated samples. Each of these straps was then subdivided into five regions, with each of the first four regions having two tensile specimens at one of the four angles. One specimen from each pair was then cut, with a razor blade knife, to create the stress raiser. The specimens were then tested on an Instron tensile machine, at the Graduate Center for Materials Research located on the University of Missouri-Rolla campus.

It was decided to give twelve other specimens a low dose of gamma radiation before any tensile tests were performed on them. These samples were taken from sections lc, $4b$, 8a, 8b, 12d, and 16c. Only 0⁰ and 90⁰ samples were used in this part of the experiment. Each of these specimens had also been exposed to a low fluence of neutrons in the attenuation experiment; however, the neutron exposure was of such a low magnitude $(< 200 \text{ n/cm}^2)$ that it was not expected to influence the results. To irradiate the samples, they were placed in a water tight plastic vial and lowered next to the reactor whenever it was shutdown. Care was taken to ensure that the samples were not exposed to any additional neutrons, so that they would not become activated. After the specimens had received approximately 5.1 MR of gamma radiation, they were removed from the pool and checked for any induced radiation on a lithium drifted germanium,

Ge(Li), detector. These samples were then tested on the Instron tensile machine.

E. TRACE ELEMENT DETERMINATION

Neutron activation analysis (NAA) is a very accurate method of identifying very small levels of impurities in a given base material without having to first use expensive chemical separation techniques [20]. The basic principle behind NAA is that an unknown material is exposed to a flux of neutrons. This causes each isotope in the material to be transformed to a new different isotope. Some of these new isotopes are radioactive, and they will give off their characteristic radiation. By placing this material on a high resolution detector, which is connected to a multichannel analyzer (MCA), an energy spectrum is able to be collected that contains the characteristic gamma radiation from each of the radioactive isotopes. This energy spectrum can then be analyzed to identify many of the isotopes that are present in the material. After the peaks are identified the average activity for each isotope can be used to calculate the mass of each element that is present in the material.

NAA is a very good technique to identify most impurities that may be present in a sample. However, there are some limitations to this method. Some isotopes have such a small cross section that they will not activate unless a very high fluence of neutrons is used. Also, even after a radioactive isotope has been produced its half life may be so long or short that it is difficult to detect them on the counting system. Some isotopes will also be hidden because of other more radioactive peaks next to them. This problem can normally be avoided by taking several counts at various time intervals, to allow the isotopes with shorter half lives to decay away, and by looking for other energy peaks that the isotope in question may emit. The final problem is that no single detector system can detect all of the different types of radiation. To avoid this limitation, several different systems can be utilized so that all emitted radiation will be detected.

The BoraflexTM was examined for trace element impurities by using neutron activation analysis. A long term irradiation was also performed on the BoraflexTM to simulate a lifetime fluence of 10^{17} n cm^{-2} . Each sample was cleaned and then sealed in a plastic fingertip, which was then encapsulated in a rabbit tube vial. This vial was inserted into the core using a pneumatic tube system. After the sample was irradiated it was placed on a Ge(Li) detector which was connected to a IBM PC based multi-channel analyzer (MCA). After the spectrum was collected it was stored on a floppy disk for further analysis. This disk was transferred to a different computer where a Gaussian fit was used to determine the location of each peak. After the peaks were found, the computer calculated the centroid energy and net activity of each peak. The computer then compared this

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energy spectrum to a library file to see if any of the peaks were identical. Finally the system gave reports over both the identified peaks and the unidentified peaks. These results were then checked by hand to determine if the identified peaks were possible, and to try to identify the remaining unidentified peaks.

F. NEUTRON ATTENUATION

1. Total Cross Section A total neutron cross section for thermal neutrons was found by using the transmission (or beam) technique. In this method a beam of neutrons from a reactor is allowed to pass through a sample and strike a detector so the neutrons may be counted. Next a count rate is taken without the sample in the beam. To ensure that only thermal neutrons are being counted, a thin sheet of cadmium is then placed in the neutron beam and an epicadmium count is taken with and without the sample present. This epi-cadmium count is a measure of the number of nonthermal neutrons that are being counted by the detector. The cadmium was chosen for this purpose because at low, or thermal, energies it has a very large absorption cross section, which suddenly drops to a much lower cross section for all higher energy levels. A transmission coefficient for thermal neutrons can then be determined using equation **3.**

12

$$
T = I / I_0 \qquad (Eq. 3.)
$$

where T is the thermal neutron transmission coefficient

- I_o is the thermal neutron count rate with no sample in place
- I is the thermal neutron count rate with the sample in place

A macroscopic total removal cross section cross section for thermal neutrons can then be calculated using equation 4 **[21, 22].**

$$
M = ln(T) / x \qquad (Eq. 4.)
$$

where M is the macroscopic total removal cross section for thermal neutrons

T is the thermal neutron transmission coefficient

x is the sample thickness

2. Absorption Cross Section The danger coefficient method was used to determine an absorption cross section for the BoraflexTM and the Al-B-TenTM. This method is an in-pile technique for measuring the reactivity of a sample. The absorption cross section can then be determined through some calculations, or by comparing the reactivity of the sample with that of a known standard [22].

To determine the reactivity of the sample, the reactor was started and taken to a low power level (20 W) and the control rod heights were recorded. The sample was then carefully inserted into the core. Then, the reactor was taken back to the same power level, and the critical rod heights were recorded again. The difference in the rod heights was then related to the reactivity of the sample through the use of the reactor's rod worth curves [22,24,25]. To simplify these results, all of the shim/ safety rods were kept at a constant height, so that the entire reactivity change was compensated by the regulating rod.

After the reactivity of the sample was determined two methods were used to calculate the absorption cross section. In the first method equation 5 was used to give an absolute value for the microscopic absorption cross section.

 $sig_a = 10^{21} * A_s * r_s / (m_s * f^2)$ (Eq. 5.)

where sig_a is the microscopic absorption cross section

- A_{g} is the atomic weight of the sample
- r_s is the reactivity of the sample (in deltaK/K)
- $m_{\rm s}$ is the mass of the sample

f is the reactor flux at 20 W

In the second method, the absorption cross section is determined by comparing the reactivity of the sample to the reactivity of a known standard. This relative method uses equation 6 to give a final value for the sample's cross section.

14

$$
sig_{a2} = r_2 * m_1 * sig_{a1} / (r_1 * m_2)
$$
 (Eq. 6.)

III. RESULTS

A. RADIOGRAPHY

The radiographs were read on a Kodak Model 1 color densitometer. The following table shows the range of density values that were found across each radiograph. The error associated with each density measurement is ± 0.05 due to the difficulty of differentiating between the light intensities in the densitometer.

TABLE 1

RADIOGRAPHIC DATA FOR BORAFLEXTM SAMPLES

B. TENSILE PROPERTIES

The tensile properties have been placed into four groups for comparison purposes. The first group was used to determine if the different load cell affected the results. The second group contains the tensile data grouped by their original horizontal location in the sheet. The third group has the tensile data grouped according to their original vertical position in the sheet, and the last group was arranged by the angular orientation of the tensile specimens to the normal direction of the original sheet of BoraflexTM.

TABLE 2

TENSILE PROPERTIES OF BORAFLEXTM GROUPED BY THE LOAD CELL USED

TENSILE PROPERTIES OF BORAFLEXTM GROUPED BY

HORIZONTAL LOCATION

Strap	Cross	Percent	Stress At
Number	Sectional	Elongation	Failure
	Area		(psi)
	(in^2)		
2	0.01611	47.3 ± 8.7	290 ± 20
$\overline{\mathbf{2}}$	0.01273	17.0 ± 3.4	190 ± 46
3	0.01611	53.3 ± 17	390 \pm 42
5	0.01611	$56.8 + 6.4$	$379 + 110$
5	0.01273	23.5 ± 3.4	268 ± 41
6	0.01611	$52.8 + 3.5$	313 ± 22
6	0.01273	18.6 ± 0.4	203 ± 15
7	0.01611	60.1 ± 4.3	$386 + 81$
7	0.01273	18.9 ± 2.2	202 ± 23
9	0.01611	55.9 ± 6.9	387 ± 105
9	0.01273	$16.8 + 1.1$	$230 + 120$
11	0.01611	85.0 ± 76	292 ± 45
11	0.01273	23.0 ± 4.1	257 ± 43
13	0.01611	51.3 ± 8.2	$311 + 40$
13	0.01273	21.2 ± 2.7	240 ± 49
15	0.01611	55.3 ± 5.1	287 ± 35
15	0.01273	22.4 ± 2.4	231 ± 26

TENSILE PROPERTIES OF BORAFLEXTM GROUPED BY VERTICAL LOCATION

TENSILE PROPERTIES OF BORAFLEX $^{\text{TM}}$ GROUPED BY

ANGLE OF SAMPLE

TENSILE PROPERTIES OF IRRADIATED BORAFLEXTM

C. TRACE ELEMENT ANALYSIS

The following tables show the concentration of all of the identified isotopes in the BoraflexTM. The first table identifies short term activation products, and the second table identifies long term activation products. The short term activation was performed at a thermal power of 20 kW for 10 minutes. For the long term activation, samples were lowered next to the core for four, four hour runs with the reactor at a thermal power of 200 kW, this gave each sample a neutron fluence of 6E+16

TRACE ELEMENTS IDENTIFIED IN SHORT TERM ACTIVATION OF BORAFLEXTM

TRACE ELEMENTS IDENTIFIED IN LONG TERM ACTIVATION OF BORAFLEXTM

D. NEUTRON ATTENUATION

Tables 8 through 10 show the results of the total removal cross section for thermal neutrons. They contain the data for BoraflexTM, BoralTM, and aluminum. The aluminum was used as a benchmark for the attenuation setup. There was a 10 percent statistical error for the attenuation data for the BoraflexTM and the Boral TM , while the aluminum only had a 2 percent statistical error in its measurement.

TOTAL THERMAL CROSS SECTION FOR BORAFLEX $^{\tt TM}$

TOTAL THERMAL CROSS SECTION FOR BORAL™

TABLE 11

TOTAL THERMAL CROSS SECTION

FOR ALUMINUM

Table 11 shows the calculated absorption cross sections for BoraflexTM, and Al-B-TenTM. This data had a 10 percent statistical error in its initial measurement.

TABLE 12

THERMAL ABSORPTION CROSS SECTION FOR BORAFLEXTM AND AL-B-TENTM

IV. DISCUSSION AND ANALYSIS OF RESULTS

A. RADIOGRAPHY

The X-ray and neutron radiographs were used to determine the uniformity of the B_4C distribution in the Boraflex TM . The two methods were also compared to see if any artifacts that were discovered using the one method could also be found by using the other. Data for the radiographs can be found in Table 1 and in Appendix B.

A penetrameter was made for each method so that the quality of the radiographs could be determined. The dimensions and corresponding film densities of these penetrameters can also be found in Appendix B. One problem that was found with the X-ray penetrameter was that even a single thickness of scotch tape or paper affected the film density. This was due to the low tube voltage and short exposure time used for the X-ray radiographs. However, the tape and paper were needed to have a consistent penetrameter. Because of this, the true density readings for the X-ray penetrameter were lower than the recorded values; however, the ratio of thickness change-to-density change is believed to be the same. The tape used for the neutron penetrameter is not believed to have caused any difference in the density readings because of the low cross section for the elements in the tape, and the high exposure time.

The results showed a fluctuation in the film density across both the X-ray and neutron radiographs of five percent. These minor fluctuations could have been because of some minor variations of the B_4C distribution or some variations in the actual polymer base material. The neutron radiographs showed some places that had a noticeable decrease in the film density. This was believed to have been caused by some of the B_4C clumping together. However, no locations were found that had noticeable density increases, which would have probably corresponded to a B_4C decrease. The average densities for all of the radiographs was very close, except for the neutron radiograph of sample 10, which had an average density of 0.82 ± 0.05 . This value can be compared to the average neutron radiograph density of 0.87 ± 0.05 . This difference corresponded to a thickness change of only 5.6%, which was not considered to be a very significant fluctuation in B_4C density. Therefore, the BoraflexTM appears to have a very uniform B_4C distribution.

The X-ray and neutron radiographs were also compared to one another to see if any artifacts would show up in both of them. The purpose of this comparison was to determine if X ray's could be used as either a supplement or substitute to the currently used neutron radiography. Samples 6 and 14 were chosen for this comparison because both of their neutron radiographs had noticeable decreases in their film density at a set location, and sample 14 had a 0.25 inch tear in one of its edges. No noticeable density changes were found in either of the X-ray radiographs; however, the tear was visible in the X-ray of sample 14. The neutron
radiograph for sample 14, on the other hand, did not show any evidence of the tear. This is due to the low cross section of the base material not having any effect on the radiograph. Even though a tear did exist in the material, it was not pulled apart so the B_4C loading was not affected. Also, the BoraflexTM is self adhesive so it did not allow the tear to gap open, unless a load was applied. Because of this type of situation, it is suggested that both X-ray and neutron radiography should be used to examine the material for defects. The neutron radiography is the superior technique to identify fluctuations in the B_4C loading, while the X-ray radiography is better at detecting flaws in the base material. X-ray radiography could also be used to determine whether a decrease in the B_4C density was due to a void in the base material, or whether it was due to a lower concentration of B_4C in the affected area.

B. TENSILE PROPERTIES

The data for the tensile properties of the BoraflexTM specimens were grouped in several manners to see what relationships, if any, existed in the material. All of the data was first sorted according to the load cell that was used by the Instron machine. The data was then sorted by the original location of the sample on the sample sheet. This included sorting by both the strap number, which was related to the original horizontal location of the specimen on the sheet, and strap section, which was related to the

original vertical location of the specimen on the sheet. The data was also grouped by angle, as measured from the normal direction of the main sheet. The irradiated samples were grouped in the same manner, with the exception that all tests were performed with the second load cell, and the only angles used were 0⁰ and 90⁰ from the normal direction of the sheet. The tensile properties of the unirradiated and irradiated samples were then compared to see if a relationship existed between any of the tensile characteristics for the material and the total gamma dose that the material had received.

From the data presented in Tables 2 through 5 and Appendix C it can be seen that no strong relationship existed between the original location of the specimen or the angle of the specimen, and its tensile properties. The best correlations were found for specimens tested at the different angles. This may be because the angular specimens were chosen from all parts of the original sheet instead of a certain horizontal or vertical location. The worst data correlation was found with the horizontally grouped data. This data had a standard deviation of 3.6% for the percent elongation and 59 psi for the tensile strength.

The average percent elongation for a smooth edged piece of material was 55 \pm 2.1. This is 2.7 times higher than the average percent elongation for the cut samples, which was 20 $+$ 1.0. The main reason for this decrease in elongation before failure was due to the fact that the material tore

very easily, and the cut induced into the side of the specimen gave a good starting point for the tear. This tear continued to increase in length, when the load was applied, until the specimen failed because of tearing. In contrast, the uncut specimens showed some necking before they failed suddenly.

The average tensile strength at failure was 303 \pm 18 psi for the uncut specimens, and 225 \pm 20 psi for the cut specimens. The failure strength was not as sensitive to the condition of the specimen as the percent elongation was. The main reason for this was that once a maximum loading was reached in the uncut samples the material experienced some necking before failure, while the cut specimens failed without any noticeable necking. This absence of any necking was what allowed the ultimate tensile strengths to be relatively close while the percent elongation results had a much larger difference.

For best results, all of the tensile tests should have been performed without any changes to the tensile machine or the room environment. Between the second and third day of testing the load cell was damaged by a different researcher, which made it necessary to replace the load cell. After the load cell was replaced, the average tensile strength was found to be lower than the value determined from the earlier data. No reason exists for this difference, since the Instron machine was calibrated before each session and the room was at a relatively constant temperature. Therefore,

the data from the first two days of tests were considered to be valid, and were therefore used to calculate the average tensile properties of the BoraflexTM.

The samples that received 5.1 MR of exposure to gamma rays had a percent elongation of 91.4 ± 8.8 and a tensile strength at failure of $144 + 14$ psi. This is a 52% lower tensile strength than the unirradiated samples, but a 66% higher percent elongation than the unirradiated samples. Previous tensile tests had been performed on the BoraflexTM material. The results from this previous test gave a tensile strength range of 151 to 263 psi for the unirradiated samples and a tensile strength range of 150 to 714 psi for the irradiated samples. The unirradiated results correspond well with the present results: however, a large decrease in the tensile properties was observed in this research for the irradiated samples, while the previous research reported a tensile strength increase. This discrepancy may be due to the amount of radiation the samples were exposed to. In the earlier research the samples were exposed to an average of 1.70E+10 R, while the current research was limited to 5.1E+06 R.

C. TRACE ELEMENT DETERMINATION

Neutron activation analysis was used to determine the amount of impurities in the BoraflexTM. All of the identified peaks are listed in Tables 6 and 7 and in Appendix D. Some other peaks were not able to be identified, either by

hand or by the computer based nuclide library. The possibility also exists for one of the identified peaks to be obscuring a peak from a different isotope. Much care was taken to prevent this situation from occurring by taking multiple counts of each sample at many different time intervals, and using different counting time increments. The isotopes that were the most likely to be missed would have to have a very short half life. Because of this, they would not give a significant contribution to any long term induced activity in the sample.

The sodium and chlorine were probably due to salt contamination in or on the fingertip, or they were extra contaminates on the outside surface of the Boraflex TM . The isotopes that have the greatest likelihood of existing in the BoraflexTM are europium, manganese, magnesium, zinc, dysprosium, and lanthanam.

D. NEUTRON ATTENUATION

A total removal cross section for thermal neutrons was determined for BoraflexTM, BoralTM, and aluminum. This cross section data can be found in Tables 9 and 10. The aluminum cross section was experimentally determined to serve as a benchmark value for the experimental setup.

The average aluminum cross section was calculated to be 0.499 barns. This is 2.14 times higher than the accepted value for aluminum (0.233 barns). Because of this result a correction factor of 2.14 \pm 2% will be utilized to calculate

the remaining transmission cross section results.

The transmission coefficients and macroscopic cross sections are believed to be the most accurate of the cross section results. There was some question as to whether the number densities used for the microscopic cross section determination were correct. For this reason, all of the results will be expressed in terms of the macroscopic cross section with the microscopic cross section enclosed in parentheses. The calculated total thermal removal cross section for BoraflexTM was determined to be 8.0 \pm 0.8 cm⁻¹ (14 \pm 1 barn), and for BoralTM it was 19 \pm 2 cm⁻¹ (220 \pm 20 barn).

The absorption cross section was found by first determining the reactivity worth of a sample of BoraflexTM and $AL-B-TenTM$, using the danger coefficient method. The cross section was then calculated by either a absolute or relative method. The absolute method used the reactivity worth of the sample and the thermal neutron flux at the irradiation position to calculate the absorption cross section. This method gave a value of $12 + 1$ barn for the BoraflexTM and a value of 4.8 \pm 0.5 barn for the Al-B-TenTM. The relative method related the reactivity change due to the unknown sample the reactivity change from a known standard. This technique gave a relative absorption cross section of 1900 \pm 200 barn for the BoraflexTM and 780 + 80 barn for the Al-B- $\texttt{Ten}^{\texttt{TM}}$. A boron standard was not available so cadmium was chosen as the standard. Because the cadmium does not have

the same absorption characteristics as the boron, this method is not expected to give exact results for the BoraflexTM and the Al-B-TenTM. However, this method did give a valid correlation factor between the absorption cross sections of $AI-B-Ten^{TM}$ and BoraflexTM. The absorption cross section of the BoraflexTM was determined to be 2.48 times higher than the absorption cross section of the Al-B-TenTM.

V. CONCLUSION

This research focused on several different properties of BoraflexTM. The radiographic studies showed a very uniform boron carbide distribution throughout the material. However, the neutron radiography showed an inability to detect tears in the material, which would decrease the materials ability to stretch before failing, and its tensile strength at rupture. This problem may not be serious since the BoraflexTM is normally clad in stainless steel, and it does not support any load. Tears could be detected if X-ray radiography would be used in addition to the normal neutron radiography. This technique is relatively inexpensive and it would be able to detect thickness fluctuations, voids, and tears in the material.

The tensile properties were checked to determine the uniformity of the mechanical characteristics of the Boraflex TM . The average tensile strength was calculated to be 303 \pm 15 psi. There was a major increase in the tensile strength for samples 5, 7, and 9. These samples had an average tensile strength at rupture of 384 ± 4.6 psi. Because of this fluctuation, sample 6 was checked to see if its properties were also high. Its tensile strength was calculated to be 317 \pm 3.2 psi. No reason is expressed for this increase; however, all of the high data was collected on the same day. Therefore, a calibration problem might have existed, even though the calibration was checked before the samples were tested. Because no reason can be found to prove this data is erroneous, it was not discarded. When the angular dependence was checked, a 10% increase in tensile strength was found for the samples at 135⁰ from the normal direction to the sheet. This may be due to a preferred orientation of B_4C grains in the polymer, or even a preferred orientation for the polymer chains. However, no reason was identified for this change in tensile strength.

Twelve tensile samples were also exposed to a gamma intensity of 5.1 MR. These specimens had a percent elongation of 91.4 \pm 8.8 and a tensile strength at failure of 144 \pm 14 psi. This is a 52% lower tensile strength than the unirradiated samples, but a 66% higher percent elongation than the unirradiated samples. Previous tensile tests had been performed on the BoraflexTM material. The results from the previous test showed an increase in tensile strength and a corresponding decrease in ductility for high levels of gamma exposure. This is the reverse of the current results. Also, some bubbles were detected on the surface of the Boraflex TM after the samples were removed from the pool. Because of these results, it is suggested for intermediate range gamma exposures be performed to give a better understanding of the materials physical property changes during a pure gamma exposure.

The trace element analysis showed the presence of europium, magnesium, manganese, zinc, dysprosium, and lanthanum. Sodium and chlorine were also identified;

although, their presence was believed to have been caused by salt contamination of the plastic fingertip or the surface of the material. Some minor peaks were not identified. However, the major detectable impurities are believed to have been identified.

Both total and absorption cross sections, for thermal neutrons, were determined in this research. The average total cross section for BoraflexTM was calculated to be 10 $+$ 1 cm^{-1} (14 \pm 1 barn). This can be compared to the total cross section for BoralTM, which was calculated to be 19 + 2 cm^{-1} (220 \pm 2 barn). The absorption cross section for the BoraflexTM and Al-B-TenTM were calculated through both a relative and absolute method. The relative method gave an absorption cross section of 1900 \pm 200 barn for BoraflexTM and 780 \pm 80 barn for the Al-B-TenTM. These results may not be exact values, though, because the standard sample used was cadmium, which has different absorption resonances than the boron. However, this technique was able to relate the absorption cross sections of the $A1-B-Ten^{TM}$ to the BoraflexTM. This relation showed that the absorption cross section for the BoraflexTM is 2.5 \pm 10% times greater than the absorption cross section for the Al-B-Ten™.

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VITA

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He has been enrolled in the Graduate School of the University of Missouri-Rolla since August 1985 and has held the United State Department of the Interior Fellowship for the period August 1985 to December 1985, and the Power Fellowship for the period January 1986 to December 1986.

APPENDIX A

ORIGINAL LOCATION OF BORAFLEX TM SAMPLES

Figure A-1. Original Location of BoraflexTM Samples

- **A Attenuation Samples**
- **R Radiography Samples**
- **T Tensile Samples**

APPENDIX B

RADIOGRAPH DATA

The following tables show the densities found at several locations on each radiograph, and the maximum discontinuity found. The specimen numbers with either a B or T are for the X-ray radiographs, and the plain numbers are for the neutron radiographs.

TABLE B-l

E1LM DENSITY EOK SAMPLE SHEET 4

FILM DENSITY FOR SAMPLE SHEET 6

No discontinuities were found.

TABLE B-3

FILM DENSITY FOR SAMPLE SHEET 10

TABLE B-4

FILM DENSITY FOR SAMPLE SHEET 14

The penetrameters for the both the neutron and X-ray radiographs were made by slicing sections of Boraflex off of an end. These pieces were as uniform in thickness as possible, however, they may not have been uniform due to the state of the material they were cut from.

TABLE B-5

X-RAY PENETRAMETER DIMENSIONS AND DATA

TABLE B-6

NEUTRON PENETRAMETER DIMENSIONS AND DATA

APPENDIX C

TENSILE TEST DATA

Figure C-l. Tensile Specimen Dimensions

- **L 5/8 + 1/64 inch**
- **LO** *lh* **+ 1/64 inch**
- **T 11/128 + 1/128 inch**
- **W 3/16 + 1/128 inch**
- **WO 5/8 + 1/64 inch**

The following tables show the results of the tensile tests.

Tensile properties grouped by the specimen's horizontal location in the Boraflex sheet:

TABLE C-l

Initial Gage Length Length (in) Final Gage (in) Percent Elongation Load at Rupture (lb) Cross Sectional Area (in^2) Stress At Failure (psi) 0.6250 0.9085 45.4 6.38 0.01611 3.96E+02 0.6250 0.9636 54.2 7.64 0.01611 4.74E+02 0.6250 1.0069 61.1 7.20 0.01611 4.47E+02 0.6250 1.0305 64.9 7.42 0.01611 4.61E+02 0.6250 0.9722 55.6 4.07 0.01611 2.53E+02 0.6250 0.9274 48.4 4.18 0.01611 2.59E+02 0.6250 0.7667 22.7 3.46 0.01273 2.72E+02 0.6250 0.7510 20.2 3.08 0.01273 2.42E+02 0.6250 0.8022 28.3 4.12 0.01273 3.24E+02 0.6250 0.7667 22.7 2.97 0.01273 2.33E+02

TENSILE PROPERTIES FOR STRAP 15

Tensile properties grouped by the specimen's vertical location in the Boraflex sheet.

TENSILE PROPERTIES FOR STRAP SECTION A

TABLE C-ll

TENSILE PROPERTIES FOR STRAP SECTION B

TENSILE PROPERTIES FOR STRAP SECTION C

TENSILE PROPERTIES FOR STRAP SECTION D

Initial	Final	Percent	Load	Cross	Stress
Gage	Gage	Elongation	at	Sectional	At
Length	Length		Rupture	Area	Failure
(in)	(in)		$(\mathbf{1}\mathbf{b})$	(in^2)	(psi)
0.6250	0.9652	54.4	4.62	0.01611	2.87E+02
0.6250	0.9833	57.3	5.88	0.01611	3.65E+02
0.6250	1.0423	66.8	7.37	0.01611	4.57E+02
0.6250	1.0030	60.5	6.27	0.01611	3.89E+02
0.6250	0.9754	56.1	5.17	0.01611	3.21E+02
0.6250	1.0305	64.9	7.42	0.01611	4.61E+02
0.6250	0.9888	58.2	5.50	0.01611	3.41E+02
0.6250	1.0187	63.0	6.54	0.01611	4.06E+02
0.6250	1.0030	60.5	7.86	0.01611	4.88E+02
0.6250	0.9942	59.1	4.51	0.01611	2.80E+02
0.6250	0.9499	52.0	5.72	0.01611	3.55E+02
0.6250	1.0090	61.4	4.84	0.01611	3.00E+02
0.6250	0.7124	14.0	2.09	0.01273	1.64E+02
0.6250	0.7667	22.7	2.97	0.01273	2.33E+02
0.6250	0.7384	18.1	2.42	0.01273	1.90E+02
0.6250	0.7352	17.6	2.42	0.01273	1.90E+02
0.6250	0.7313	17.0	2.20	0.01273	1.73E+02
0.6250	0.7579	21.3	2.75	0.01273	2.16E+02
0.6250	0.7727	23.6	3.74	0.01273	2.94E+02
0.6250	0.7727	23.6	2.86	0.01273	2.25E+02

TENSILE PROPERTIES FOR STRAP SECTION E

Tensile properties grouped by the angle from the normal direction of the Boraflex sheet.

TENSILE PROPERTIES FOR SPECIMEN AT 0 DEGREES

TENSILE PROPERTIES FOR SPECIMEN AT 0 DEGREES

TENSILE PROPERTIES FOR SPECIMEN AT 45 DEGREES

TENSILE PROPERTIES FOR SPECIMEN AT 90 DEGREES

TENSILE PROPERTIES FOR SPECIMEN AT 135 DEGREES

TABLE C—19

TENSILE PROPERTIES FOR SPECIMENS TESTED WITH

TABLE C-19 --CONTINUED

TENSILE PROPERTIES FOR SPECIMENS TESTED WITH

TENSILE PROPERTIES FOR SPECIMENS TESTED WITH

TABLE C-20 --CONTINUED

TENSILE PROPERTIES FOR SPECIMENS TESTED WITH

TABLE C-20 --CONTINUED

TENSILE PROPERTIES FOR SPECIMENS TESTED WITH

TENSILE PROPERTIES FOR IRRADIATED SPECIMENS

APPENDIX D

TRACE ELEMENT ANALYSIS DATA

The following tables contain the data for the short term trace element analysis.

TABLE D—1

IDENTIFIED ISOTOPES FOR BORAFLEX™ FOR SHORT TERM ACTIVATION

TABLE D-2

IDENTIFIED ISOTOPES FOR BORAFLEX™ FOR LONG TERM ACTIVATION

APPENDIX E

NEUTRON ATTENUATION DATA

Neutron attenuation data derived from the transmission method.

TABLE E-l

TOTAL THERMAL CROSS SECTION FOR BORAFLEXTM

TABLE E-2

TOTAL THERMAL NEUTRON CROSS SECTION FOR BORALTM

TABLE E-3

TOTAL THERMAL NEUTRON CROSS SECTION FOR ALUMINUM

Neutron attenuation data derived from the danger coefficient method.

TABLE E-4

THERMAL ABSORPTION CROSS SECTION DATA FOR BORAFLEXTM, AL-B-TENTM, AND CADMIUM

