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## THE PURIFICATION OF MISSOURI FIRE CLAYS

By

### JAMES OFFUTT

A

### THESIS

Submitted To The Faculty Of The

SCHOOL OF MINES AND METALLURGY OF THE UNIVERSITY OF MISSOURI in partial fulfillment of the work required for the

## Degree Of

BACHELOR OF SCIENCE IN CERAMIC ENGINEERING

Rolla, Missouri

1932

#### INTRODUCTION

At the present time there appears to be an unlimited supply of high grade fire clay available for most any purpese. But some time in the future there will rise the problem of purifying the low grade clay to such an extent that it will be used in high grade fire clay products.

Whatever purification method may be employed it must be applicable on a commercial scale and the cost of purification must be very low.

Until the present time little if any work along this line has been reported. Apparently no one has ever attempted and serious efforts to purify low grade fire clay. Therefore, the field is too broad to cover in a short time. It must be developed step by step.

It is the object of this research to pursue several possible and plausable methods of purification and to determine the approximate merits of each in such a manner that latter research men may know which fields show the most promise. With this object in mind we proceed.

#### REVIEW OF PREVIOUS WORK

So far as could be determined from a careful survey of the Journals of the American Ceramic Society and other literature, very little work has ever been done along the line of purification of fire clays. It is apparent that the majority of manufactors consider the present supply of fire clay to be almost inexhaustable. Some of the methods used in the furification of other clays could not be used at all with fire clay.

However, Mr A.V. Bleininger included fire clay with a number of other clays in an attempt to add to the good qualities of the clay by electrical separation. His report was written for the Transactions of the American Ceramic Society. This report was printed in the 15th volume of the Transactions issued in 1913.

An abstract of the article as far as it pretains to fire clay follows:

A very thin slip was made of the material being studied. It is w well known fact that all of the particles are supposed to carry charges(electrical). When electrodes are placed in the slip the particles will seek one or the other of the electrodes.

The fire clay was ground through 40 mesh and made into a thin slip. The fire clay used was Number 3 Ohio Fire Clay. About 0.1% caustic soda was added to the slip to act as an electrolyte.

The tests of the final clay obtained gave very good res-

ults. The tests made a very good appearance with a very uniform color including no iron spots. The shrinkage was reduced appreciablly.

This test apparently has no primary part in the purification of fire clays. The process simply replaces the filter press to a certain extent. The operation is very simple and this is a decided advantage. The writer mentions that there was no hope at that time of ever being able to settle ferric iron out of the slip. A follow up of the method was to be made, according to the writer, but apparantly nothing was ever done to advance the work since no further reports could be found.

This method of purification may work very well for some clays but the results obtained with fire clay were not of such a nature to warrant electrical separation on a commercial scale.

### RAW MATERIALS

Cone 31 has been set by leading Ceramists of the country as the dividing line betwwen low and high grade fire clays. It is apparent that any clay which had a pyrometric cone equivalent of more than 31 whould demand a higher sale price.

Any cheap method of trnasfixing a lower grade clay to such and extent that its P.C.E. value whuld be cone 31 or higher would be profitable.

In open pit mining there is always much clay of a lower quality than can be used for number one fire brick. In most open pit plants there is also a huge dump of very low grade clay filled with many impurities. Both of these conditions prevail at the A.P. Green Fire Brick Plant at Mexico, Missouri.

Samples of low grade material from the pit and a representative sample of the dump clay were used in this research.

#### DISCUSSION OF PROCEDURE

Several possible methods of purifying the fire clay were pursued in this research. A breif outline of each method is given below:

### Crigional samples

A uniform sample of the origional material was crushed in the rolls through 20 mesh and moulded stiffmud into standard test bars. Test cones were moulded from the material.

#### Disintegration

A representative sample of the origional materials was# pass through a squirrel cage disintegrator. The sample from the dump was divided into two portions one passed through 40 mesh and the other through a 20 mesh screen. All of the pit sample was passed through the 20 mesh screen. Standard test bars and test cones were moulded stiff mud from the materials.

#### Washing

A good quantity of the material was placed in a blunger and allowed to slake until nearly all of the material was in the form of a slip. The slip was then divided into four which were proportions is screed respectively in the slip state through 40, 80, 100, and 150 mesh screeks. The slip was allowed to dry sufficiently to mould stiff mud into test bars and test cones.

### Weathering

Samples were placed in perforated pans, covered with thin cloths and set on the roof of a building to be summitted to the weather. At the end of one and two months respectively a sample of the material was crushed through 20 mesh moulded stiff mud into t st bars and test cones.

#### TESTING THE BARS

The bars were fried bone dry in an electric drier and fired to cone 19 in a test furnace by following the firing schedule give elsewhere in this report.

Green shrinkage, fired shrinkage, green modulus, fired modulus, absorption, and P.C.E. values were determined for each group. The P.C.E? tests were run in a Fulton-Coursen test furnace.

#### LEGEND

ODOR--Origional sample from dump. OPCR--Crigional sample from pit. CPW1--Pit sample weathered about one month. ODW1 -- Dump sample weathered about one month. OPW2 -- Pit sample weathered two months. ODW2--Dump sample weathered two months. CPSL40--Pit sample slaked and wet screened thru 40 mesh. 11 Ħ n 11 Ħ 11 OPSL80-- " 80 Ħ \*\* OPSL100--" 11 Ħ 11 11 Ħ Ħ 100 Ħ OPSL150--" Ħ Ħ 11 . Ħ 150 Ħ n tt Ħ 11 n Ħ 40 11 CDSL40--Dump ODSL80-- " 11 Ħ 11 Ħ Ħ Ħ 80 11 11 ŧ Ħ Ħ ODSL100--" Ħ 11 Ħ 100 N 11 Ħ tt Ħ ODSL150--" 11 150 11 OPDIS40 -- Pit sample disintegrated and screened thru 40 mesh, ODDIS20 -- Dump sample disintegrated and screened thru 20 mesh. ODDIS40--Ħ . Ħ Ħ = 40 11 11

# SUMMARIZED DATA

Bars	Drying Shrk.	Fired Shrk.	Green Mod.of Rupt.	Fired Mod.of Rupt.	Abs. P	.C.E.	Comments
ODOR	8.4	4.2	24	2355	•23		light tannish yellow-iron spots.
CPOR	4.9	4.0	0	1760	5.3		Yellowish brown Iron speckeled
o <b>PW1</b>	5.3	4.6	40	1410	7.1		Yellowish tan Few bad iron spots
ODW1	9.6	3.8	26	1810		s than 30	Dark brown Few iron spots
0 <b>7</b> ₩2	4.4	4.1	4 <b>4</b>	1260	5.9		Light yellow tan Iron spots
ODWŻ	9	6.2	228	2430	.81	31	Uniform d <b>ar</b> k brown
OPSL40	9.6	8.5	167	2440	.62 30-3	31	Chocolate brown
OPSL80	9.7	9.0	162	2550	0 30-	31	Uniform greenish brown
CPSL100	9.4	9.2	190	2320	0 Less t		Uniform dark green Black cored
OPSL150	9.2	9.0	180	1840	о и		99 <del>19</del> 19
ODSL40	13.3		226		Less ti 30		All ODSL bars were
CDSL80	13.1		5 <b>60</b>				overfired, bloated cracked, swollen,
ODSL100	14.0		364		98		and dis-colored. ODSL100 was laminated
0D <b>SL</b> 150	13.9		610		<b>88</b>		somewhat.
OPDIS40	5.8	5.5	0	1220	9 31-3		Brown tan Few iron spots
ODDIS20	6.5	5.5	104	2920	1.2 31-3	2	Light tan
GDD1840	8.1	5,9	95	3420	0 31-3		Ligth speckled tan

# LEGEND OF SYMBOLS USED IN DATA

G.LGreen length of bar given in inches.
D.LDry length of bar given in inches.
F.L. Fired length of bar given in inches.
D.SDry shrinkage of bar given as a percentage of the dry length.
F.SFired shrinkage of bar given as a percentage of the dry length.
B Breadth of break in the modulus of rupture test.
D. HyDepth of break in the modulus of rupture test.
LLength of span in the modulus of rupture test. (Note: D <sub>g</sub> B, and L are in inches.)
G.M.RGreen modulus of rupture in pounds per square inch.
F.M.RFired modulus of rupture in pounds per square inch.
D.WDry weight of bar in absorption test.
W.WWet weight of bar in absorption test.
Abs Absorption in percentage with reference to the dry weight.
P.C.EPyrometric cone equivalent.

# Data Sheets

Bar -	G.L	<b>D.L.</b> -	F.L	D.S.	-F.S.	<u>-B</u>	D	Lo.	G.M.R.	-F.M.R.	-D.W.	-W.W	ABS.
ODOR 1	5	4.60	4.48	8.7	2.6	.86	.90	200		2160	75	<b>7</b> 5	0
2	5	4.62	4.36	8.2	5.6		.87			2300	72	72.5	
2 3	5	4.62	4.42	8.2	4.3		.88			2260	12	12.00	•00
4	5	4.00	1010	0.2	4.0			2407	230	~~~~			
5	5						.85		200	2700	75.5	75.5	0
Average				8.4	4.2			200	230	2355	1010		.23
OPOR													
1	5												
2	5												
3	5	4.75	4.56	5.2	4.0	.94	.85	160		1760	94	99	5.3
4	5												
5	5	4.78		4.6					0				
Averag	e 5			4.9	4.0				0	1760			5.3
CPW1		i dina alian di kata dina dina dina di					***			. Constant (1999) (1999) (1999) (1999)			
1	5	4.76	4.56	5.1	4.2	.95	•88	110		1010	82	88	7.3
2	5	4.72	4.48	5.9	5.1	.93	.85	135		1500	84	90	7.1
3	5	4.77		4.8									
4	5					.98	•92	4.5	40				
5	5					.93	.89	170		1730	86.5	92.5	6.9
Averag	e			5.3	4.6				40	1410			7.1
ODW1													
1	5	4.57	4.40		3.7		• <b>8</b> 5			<b>2</b> 120	65		.72
2	5	4.56	4.40	9.6	3.5	.94	•85	130		1500	68.5		.73
3	5	4.57	4.38	9.6	4.2		.86			2300	70.5	80.5	0
4	5				19-22	•94	•90	26.3	260				
5	5					• 92	•86	120		1320	73	73.5	•68
Averag	8			9.6	3.8	فاسترك النوار مرور البوار			260	1810		<b></b>	• 53

# Data Sheets

Bar -	G.I.	<u>- B.L.</u>	- F.L D.	SF.SBDLo G	.M.R.	F.M.R	D.WW.W.2	ABS.
OPW2 1	5	4.79	4.63 4.	4 3.4 1.02 .96 105		840	100 105	5
2	5	4.79		\$\$0\$\$\$\$				
2	C	2010		4.4 1.00 1.00 170		1220	93 99.5	6.9
3	5	4.79	4.58 4.4	4.4 1.04 1.00 180		1430	91 96	5.5
4	5			1.12 1.04 741	44			
5	5			1.02 .98 205		1560		
Averag	e		4.4	4.1	44	1260		6.5
ODW2 1	5	5.05	4.749	6.1 1.0 .99 280		2160		
2	5	5.05	4.759	6.1 1.0 .95 370		3100	105 105.5	•47
3	5	5.05	4.72 -19	6.4 1.06 1.03 34.3	228			
4	5	5.05	4.729	6.4 1.00 .98 260		2020	93 94	1.08
5	5			1.00 .97 310		2440	123.5 124.	5 .89
Averag	e		9	6.2	228	2430		.81
OPSL40 1	5	4.55	4.13 9.9	7.1 .87 .80 270		3640	94 95	1.05
2	5	4.59	4.15 9.0			1510	76.5 77	•64
3	5	4.55	4.15 9.9			2800	72 72	0
4	5			1.04 .96 21.2	167			
5	5			.90 .88 170		1820	65.5 66	.77
Averag			9.6	8.5	167	2440		.62
OPSI-80				<b>T A A A A A A A A A A</b>	<u></u>	0700	0.6 9.6	0
מנ	5	4.54	4.18 10.1			2380	86 86	
2	5	4.55	4.12 9.9			1930	70 70	0
3	5	4.58	4.18 9.2			3350	96 96	0
4	5			1.03 .95 20	162			
5			~ ~		160	2550		0
Averag	<b>.</b>		9.7	9.0	162	2000		v

# Data Sheets

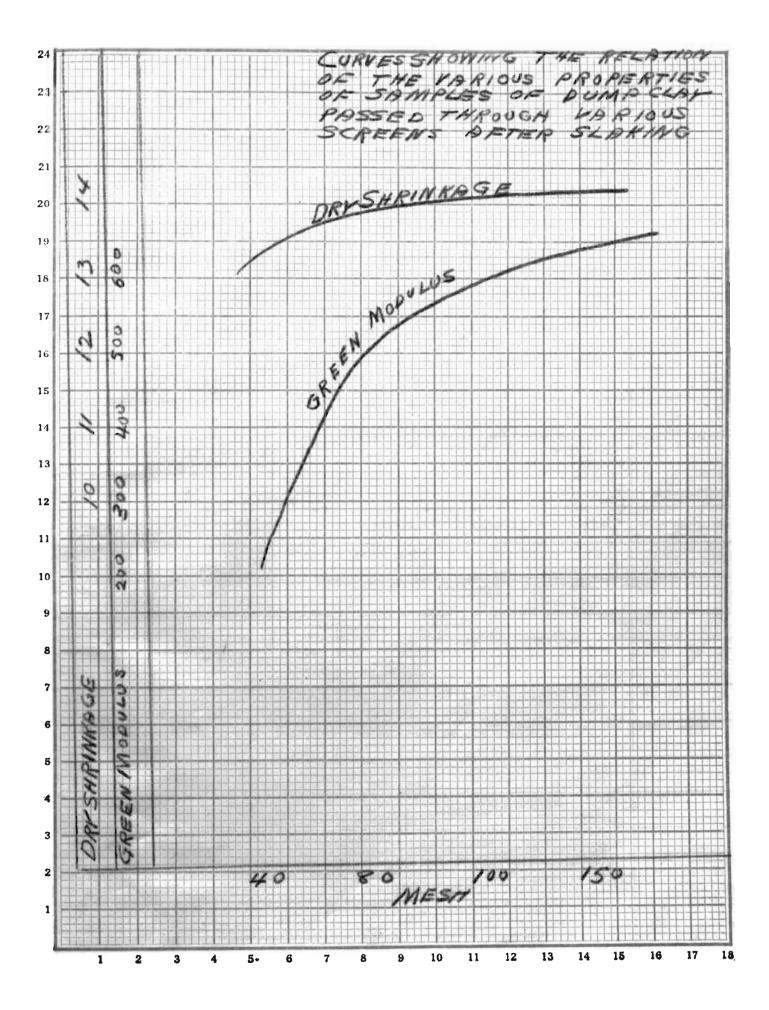
Bar -	G.L.	<b>B.L</b> .	- F.L.	- D.S	_F.8	з. <u>-</u> В.	-D	Lo .	<b>B.</b> M.R.	-F.M.R	D.W.	-W.W,	ABS.
OPSLIC 1	5	4.57	4.15	9.4	9.2	.93	.82	250		3000	86	86	0
2	5	4.58	4.18	8.7	9.4	.95	.82	140		1650	72	72	0
3	5	4.58	4.14	9.4	9.6								
4	5					1.06	•98	25.6	190				
5													
Avera	ge			9.4	9.2				190	2320			0
OPSL18	50 5	4.58	4.18	9.2	8.7	.92	.83	230		2720	95	95	0
2	5	4.59	4.15	9.3	9.6	.94	.85	105		1160	<b>9</b> 5	95	0
3	5	4.57	4.18	9.2	8.7	•94	.89	190		1920	81.5	81.5	0
4	5					1.06	<b>.9</b> 5	23	180				
5	5					.95	.86	150		1580	73	73	0
Avera	ge			9.2	9.0				180	1840			0
DSI4													
1	5	4.42		13.1									
2	5	4.40		13.6									
3	5	4.41		13.4									
4	5					.92	.87	20.9	226				
5													
Avera	ge			13.3					226				
ODSL80 1	) 5	4.42		13.1	· · · · ·				in-sijker dan same dini sider i dini sider		Maran (1997) - Alexandrian (1997) - Alexandria		
2	5	4.42		13.1									
3	5	4.42		13.1									
4						.92	.87	52.4	560				
5													
Averag	çe			13.1					560				

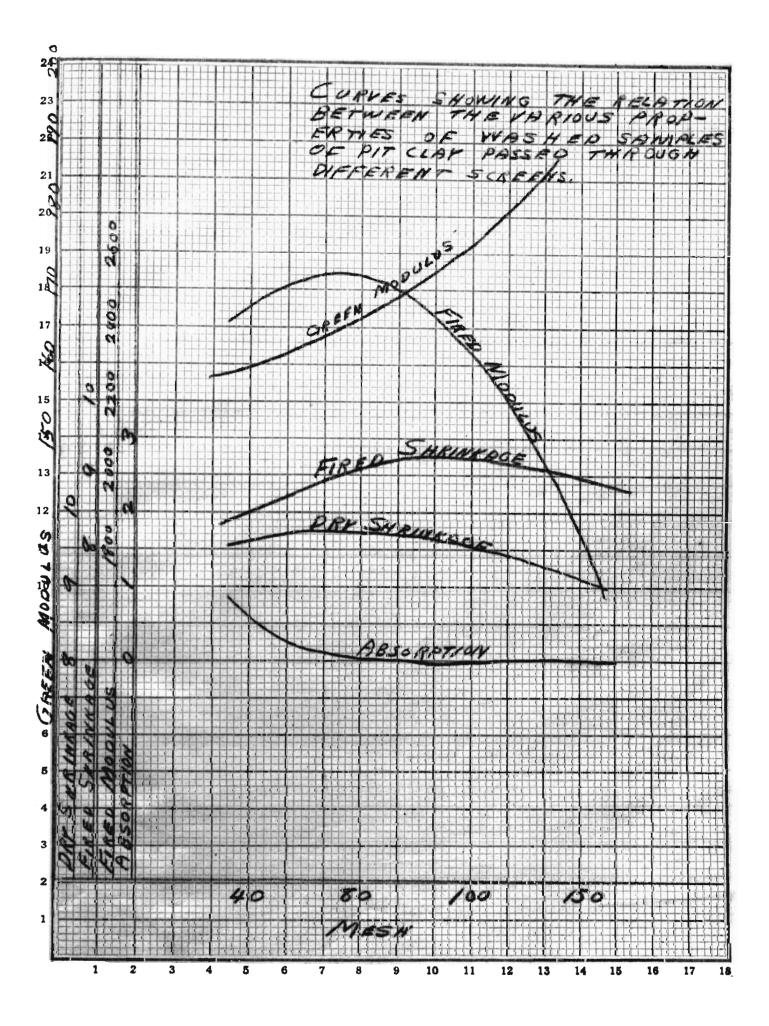
# DataSheets

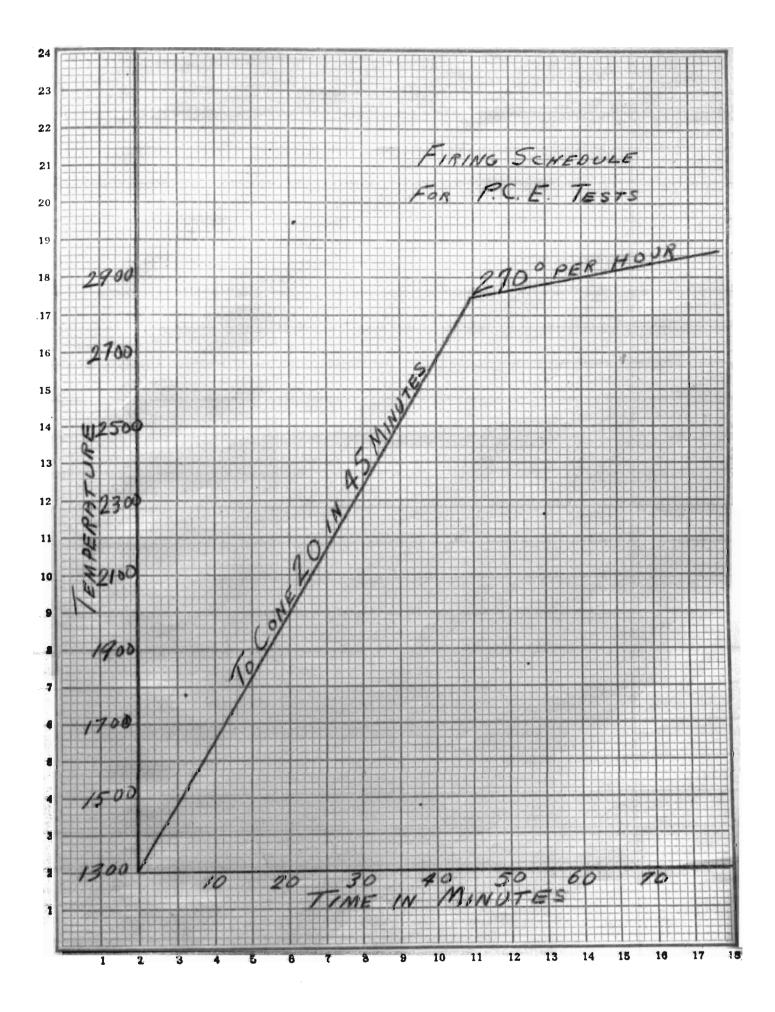
3 <b>5 4.64</b>	4.35 7.	8 6.3	.90 .84	305	3600	
4 5			.95 .91	10 95	5	

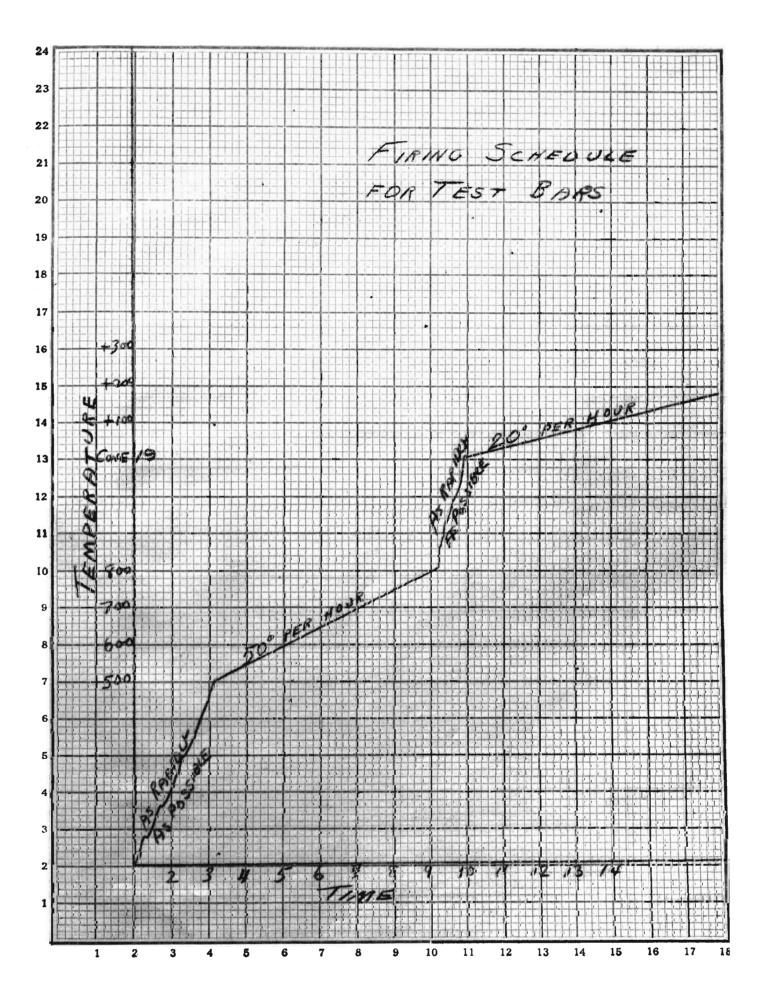
# DataSheets

Bar -	<b>D</b> .L.	<u>-</u> D.L.	- F.L.	- D.S	F.S	B	D.	-Lo	G.M.R	F.M.R.	-D.W.	-W.W	ABS.
ODSL10 1	5	4.40		13.	6								
2	5	4.38		14.	2								
3	5	4.38		14.	2								
4	5					• 90	.8	3 30	364				
5													
Averag	е			14.	0				364				
ODSL15	0 5	4.39		13.	e								
2	5	4.39		13.									
2 3	5	4.37		13.									
4	5			740	0	. 80	2.8	3 50	610				
5	U					•01		5 00	010				
Averag	<u>م</u>			14.	٦				610				
-				• * •	*								
OPDIS4	0 5	4.66	4.42	7.3	5.2	.86	.86	110		1300	89	97	9
2	5	4.78	4.50	4.6	5.9	.88	.86	100		1140	78	85	9
3	5	4.74		5.5									
4	5												
5													
Averag	8			5.8	5.5				0	1220			9
ODDI82	0	4.68	4.44	6.8	5.2	.91	.88	220		2350	62.5	71	2.1
2	5	4.68			5.6		.86			3340	88	88.5	.57
3	5		4.44					260		3020	78	78	0
4	5							11.2	104	-			
5	5							270		<b>296</b> 0	72	73.5	2.1
Averag				6.5	5.5	- • -	-		104				1.2
				te the second	و مورد او بروز و من								









#### Discussion of Data

The firing schedule used for the tests was evidently too fast and the bars showed signs of overfiring in nearly every case. The schedule allowed for volatilation and soaking of the heat mearthe maturing temperature. In the case of the washed dump clay the bars were so badly broken down that no tests at all could be obtained. The bars wheih were made from the disintegrated material sh wed the least signs of overfiring. This indicates that the quality of these bars is superior to the remainder. However, the maturing temperature was too high and much of thedata will be distorted.

The test cones for P.C.E. determination were all made in the same manner but much trouble was encountered with the furnace. Indicative results were obtained and this was all that was desired. In the P.C.E. was below cone 30 no definite value was determined.

The longer weathering tests indicated that the P.C.E. might be raised somewhat by an extended weathering period . The pit samples showed much promise along this line although the dump samples showed very little signs of improvement.

In the washed materials the cone equivalent was fairly good for the coarser meshes but the finer mesh material gave a much lower fusion point. Washing lowered the fusion point in nearly all cases and so washing may help the clay in some mays but not so far as the fusion point is concerned. The disintegrated material showed the most promise. The quality of the material was greatly increased. The stren@ gth was good and the P.C.E. was cone 32. The indicates the reason for the little overfiring on the part of these bars.

#### shrinkage

Drying/is relative but it is interesting to note that in the **case** of the dump clay after one months of weathering about 10 percent drying shrinkage was evident while after two months weathering the same clay showed 1 percent expansion on drying.

The washed clays alone gave excessively high drying shrinkage Some of them ranged as high as 13 pe cent which is too high for fire clays.

The green modulus is important only to a limited degree. Only the origional pit sample and thed is integrated pit sample gave too low green modulus for good fire clay.

The fired modulus dropped in the weathered material as the weathering progressed. This indicates that some of the fluxes were washed out of the material. The washed samples showed a higher modulus but the material through the finer meshes tended to drop indicating that the fluxes present were of a fairly coarse texture. The disintegrated material developed a wonderful fired modulus.

The importance of absorption depends on the use for which the ware is intended. The absorptions in thes group ranged from 0 to 9 percent as shown on the data sheets.

The accompanying curves are self explanatory.

#### CONCLUSIONS

Although the tests in this research were not complete or entirely conclusive it is evident that the dump clay is valuable in many respects and the pit clay has excellent changes of developing into a high grade fire clay.

All of the methods of purification pursued gave a better clay although in many cases the clay was not developed into a better or higher grade fire clay. Some of the purified products would work excellently in some other phase of the ceramic industry.

Of all of the methods employed the disintegration of the dump sample gave by far the best results. This method of purification made the necessary change in P.C.E. value from below to above cone 31 and at the same time developed a low absorption and a high strength both in the green and the fired state.

The writer recommends this method for further research in an effort to develow low grade fire clays.