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INVESTIGATIONS ON OKLAHOMA CHATS

By

Homer Chalmers Kerr

and

August Francis Delaloye

A

THESIS

Submitted to the faculty of

THE SCHOOL OF MINES AND METALLURGY OF THE UNIVERSITY OF MISSOURI

in partial fulfillment of the work required for the

D E G R E E O F

BACHELOR OF SCIENCE IN MINE ENGINEERING

and

BACHELOR OF SCIENCE IN MINE ENGINEERING

Rolla, Mo.

1921

Approved by


Professor of Metallurgy and Ore Dressing.

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INVESTIGATIONS ON OKLAHOMA CHATS

Introduction

That the tailings piles of the Joplin and the Oklahoma zinc mining fields contain zinc in commercial quantities is a known fact. Realizing this fact, a number of experiments were performed to determine whether or not this zinc could be extracted.

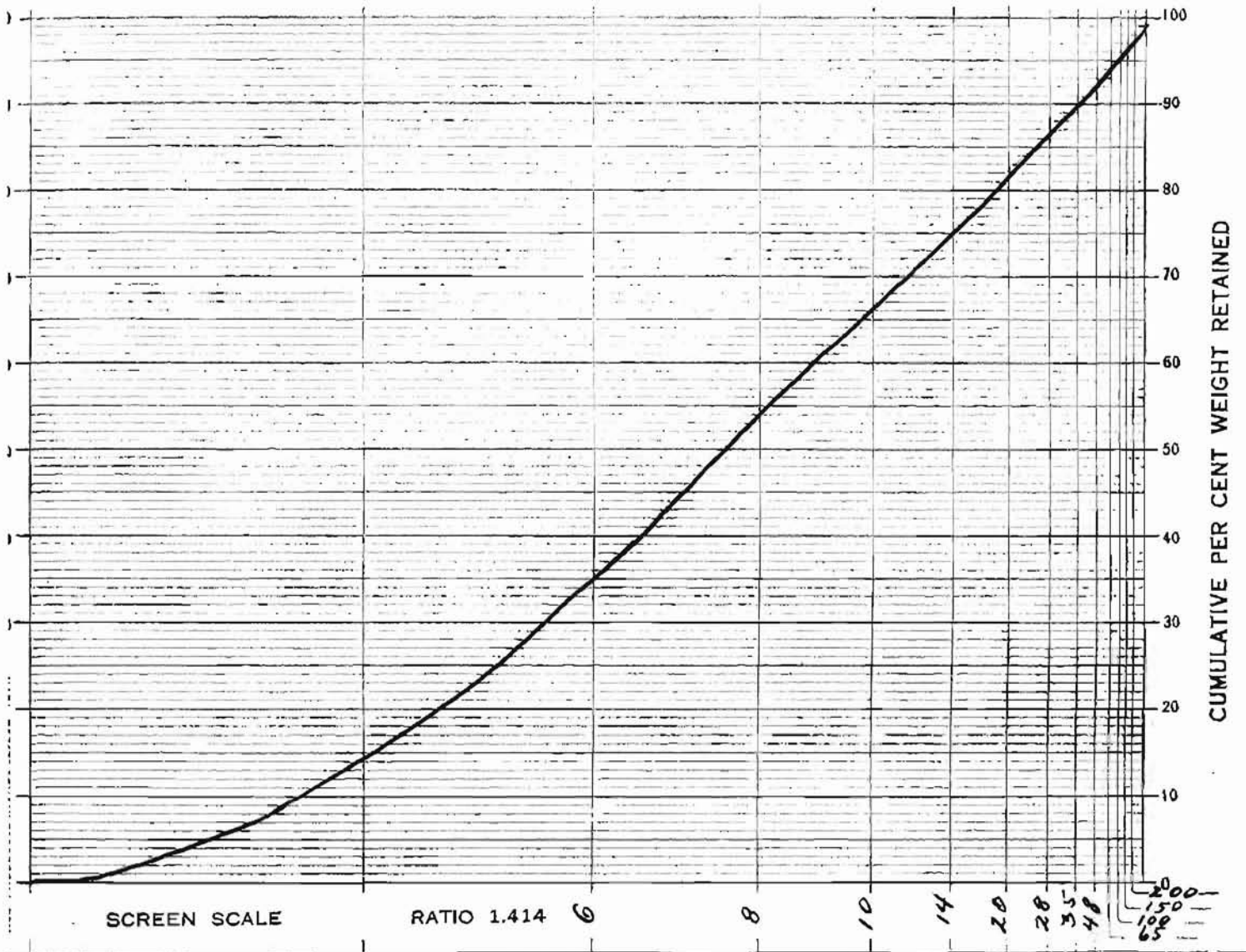
The chats for this work were obtained from the Fort Worth Mill at Pitcher, Oklahoma. They were received in two lots, one being a sample of the tailings from the rougher jigs, and the other a sample of the tailings from the tables. The chats from the rougher jigs were sampled and assayed. The assay showed them to contain 2.85% zinc, 5.92% iron, 1.40% lime, 80.0% insoluble.

A screen analysis of these chats is shown on page 3. By studying the material which remained on each screen, it was found that all the blende was broken free when the material passed a 65-mesh screen. It was then decided to find out what per cent, by weight, of free gangue would be found on the larger screens. By actually separating the grains of free gangue from the blende and the particles which contained some zinc, results were arrived at as shown on the screen analysis on page 3. From this screen analysis, one can see that 76.4% of the total weights of the chats remained on the 14-mesh screen. By combining these results, it was found that 55% of the chats on screen No. 14 contained no zinc. After arriving at these re-

sults an attempt was made to use the jigs in order that a clean tailing might be obtained, thus eliminating much of the material before fine grinding.

The Tyler Standard Screen Scale

Cumulative Direct Diagram of Screen Analysis on Sample of Material as received
 Name _____ Date _____



to the Screen bed through (also First Retaining Screen)		SCREEN SCALE RATIO 1.414				Sample Weights	Per Cent	Per Cent Cumulative Weights	%	%	%	%
		Inches	Milli-meters	Mesh	Diameter Wire Inches							
	1.050	26.67		.149								
	.742	18.86		.135								
	.526	13.33		.105								
	.371	9.423		.092								
	.263	6.680	3	.070								
	.185	4.699	4	.065								
<i>on</i>	.131	3.327	6	.036	109.2	36.6	36.6	224	5.2		90	
	.093	2.362	8	.032	50.0	16.2	52.8	2.25	5.3		70	
	.065	1.651	10	.035	42.5	14.3	67.1	2.51	4.3		70	
	.046	1.168	14	.025	27.6	9.8	76.9	2.95	4.4		62	
	.0328	.833	20	.0172	15.5	5.2	81.6	2.35	4.2			
	.0232	.589	28	.0125	14.9	5.0	86.6	2.03	3.6			
	.0164	.417	36	.0122	9.1	3.1	89.7	1.81	3.6			
	.0116	.295	48	.0092	6.1	2.1	91.8	1.81	3.8			
	.0082	.206	65	.0072	4.2	1.5	93.3	1.87	3.0			
	.0058	.147	100	.0042	2.6	1.3	94.6	2.89	.4			
	.0041	.104	150	.0028	2.7	1.2	95.8	5.81	.5			
<i>Thru</i>	.0029	.074	200	.0021	1.8	0.9	96.7	9.65	.5			
	.0026	.074	200	.0021	8.4	2.9	99.6	15.88	1.8			
					Totals	295.7	99.6					

Experiment One

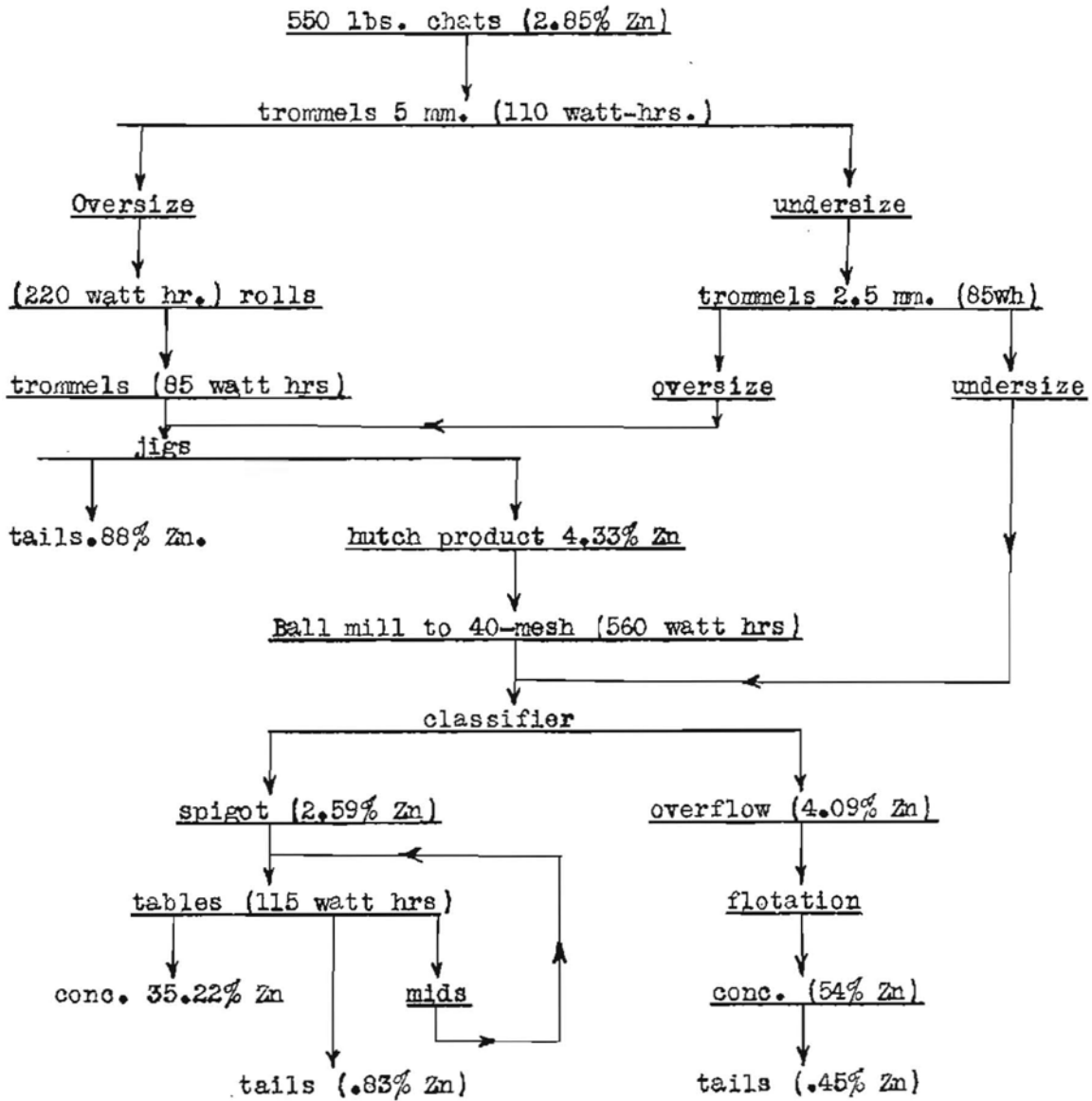
For the first run 550 lbs. of chats was weighed up and trommeled. Two revolving trommels were used, one having a 5 mm. opening and the other a 2.5 mm. opening. The material which remained on the 5 mm. trommel was passed through fine rolls and retrommeled. The material which then remained on the 5 mm. trommel was jigged with the material which passed through the 5 mm. trommel and that which remained on the 2.5 mm. trommel.

No attempt was made to obtain a concentrate from the jigs, the idea being simply to obtain a clean tailing. The material from the hutch was then ground in the ball mill and classified in the hydraulic classifier. The material which passed through the 2.5 mm. trommel was classified through the same classifier as the material obtained from the ball mill. Two products resulted; namely, a spigot product and an overflow product. Concentration of the spigot product was obtained by means of tables, and that of the overflow by means of flotation. A flow sheet with results for this method will be found on page 5.

These figures give the extraction of the material which was tailed as 52% and that which was floated 95%, or a total extraction of 78%. This figure would give a total of 12.23 lbs. of zinc recovered from the 550 lbs. of chats treated.

Estimating the cost of electrical power to be \$0.02 per KW hour, the total cost for power would be \$0.037 per 550 lbs. treated, or \$0.1245 per ton of chats treated. This would make the cost of power equal to .302 cents per pound of zinc recovered.

Flow Sheet for Experiment No. One.

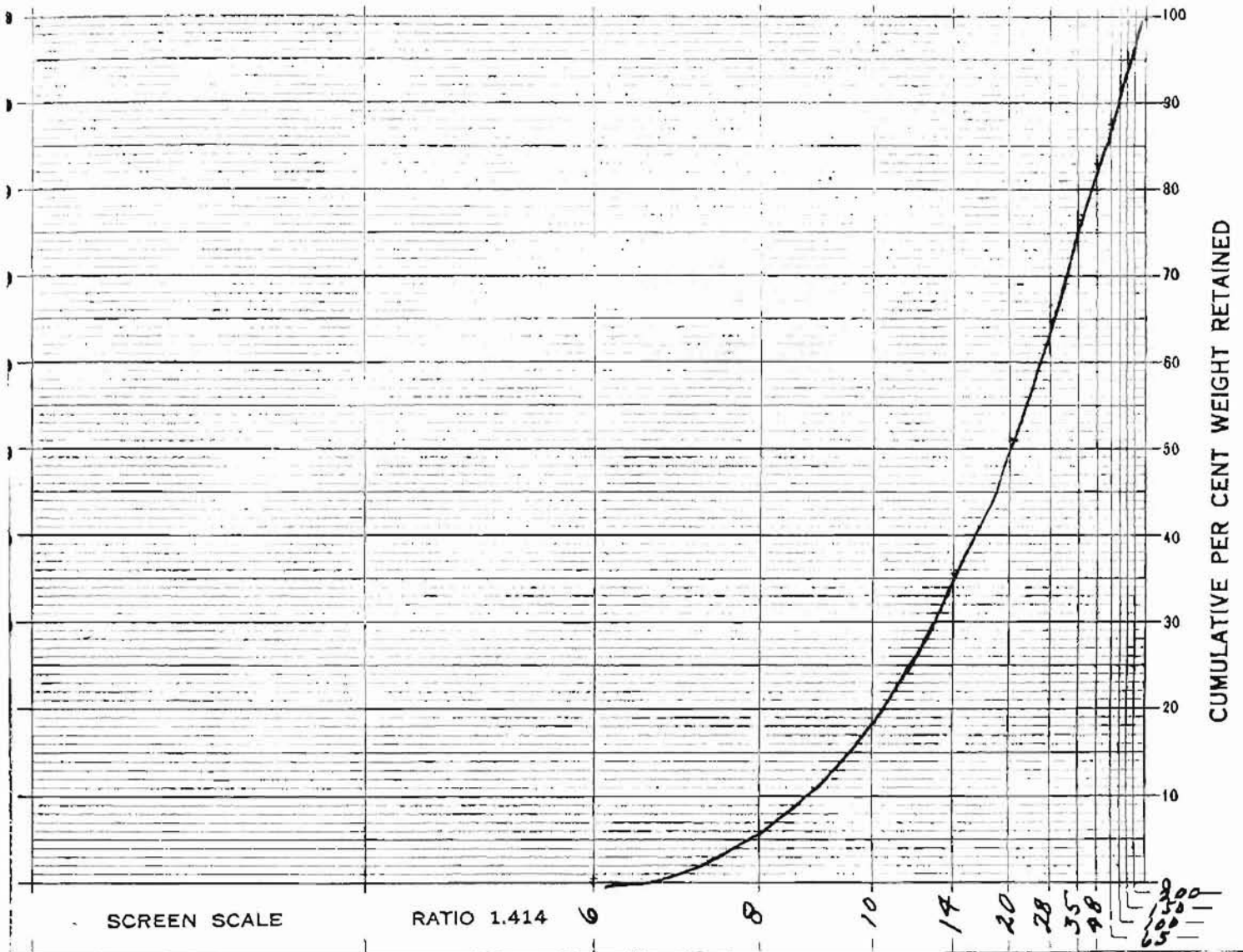


The Tyler Standard Screen Scale

Cumulative Direct Diagram of Screen Analysis on Sample of Material from rolls.

Name _____

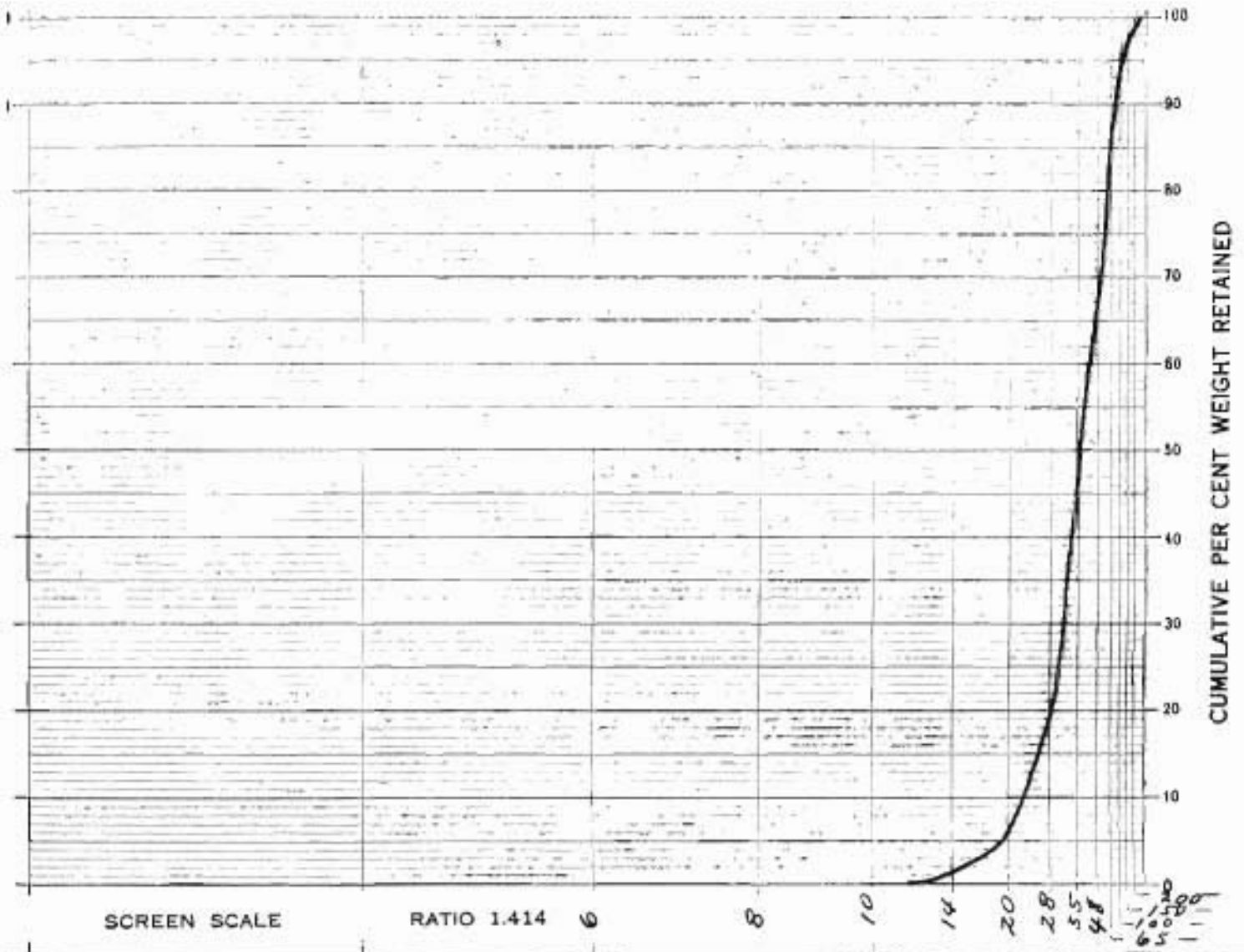
Date _____



		SCREEN SCALE RATIO 1.414										
		Openings		Mesh	Diameter Wire Inches	Sample Weights	Per Cent	Per Cent Cumulative Weights				
		Inches	Milli-meters									
		1.050	26.67		.149	Gms.						
		.742	18.86		.136							
		.526	13.33		.105	300 gm sample						
		.371	9.423		.092							
		.263	6.680	3	.070							
		.185	4.699	4	.065							
	ON	.181	3.327	6	.060	1.90	0.6	0.6				
	"	.093	2.382	8	.032	13.90	4.6	5.2				
		.066	1.661	10	.035	40.35	13.4	18.6				
		.048	1.168	14	.025	50.63	16.8	35.4				
		.0328	.833	20	.0172	46.70	15.6	51.0				
		.0232	.589	28	.0126	46.73	15.6	66.6				
		.0164	.417	35	.0122	29.40	9.8	76.4				
		.0116	.295	48	.0092	20.05	6.6	83.0				
		.0082	.208	65	.0072	14.24	4.7	87.7				
		.0058	.147	100	.0042	11.75	3.9	91.6				
		.0041	.104	150	.0028	8.92	2.9	94.5				
		.0029	.074	200	.0021	3.80	1.2	95.7				
	THRU	.0020	.074	200	.0021	11.60	3.8	99.5				
				Totals		299.97	99.5					

The Tyler Standard Screen Scale

Cumulative Direct Diagram of Screen Analysis on Sample of Spigot Product
 Name _____ Date _____



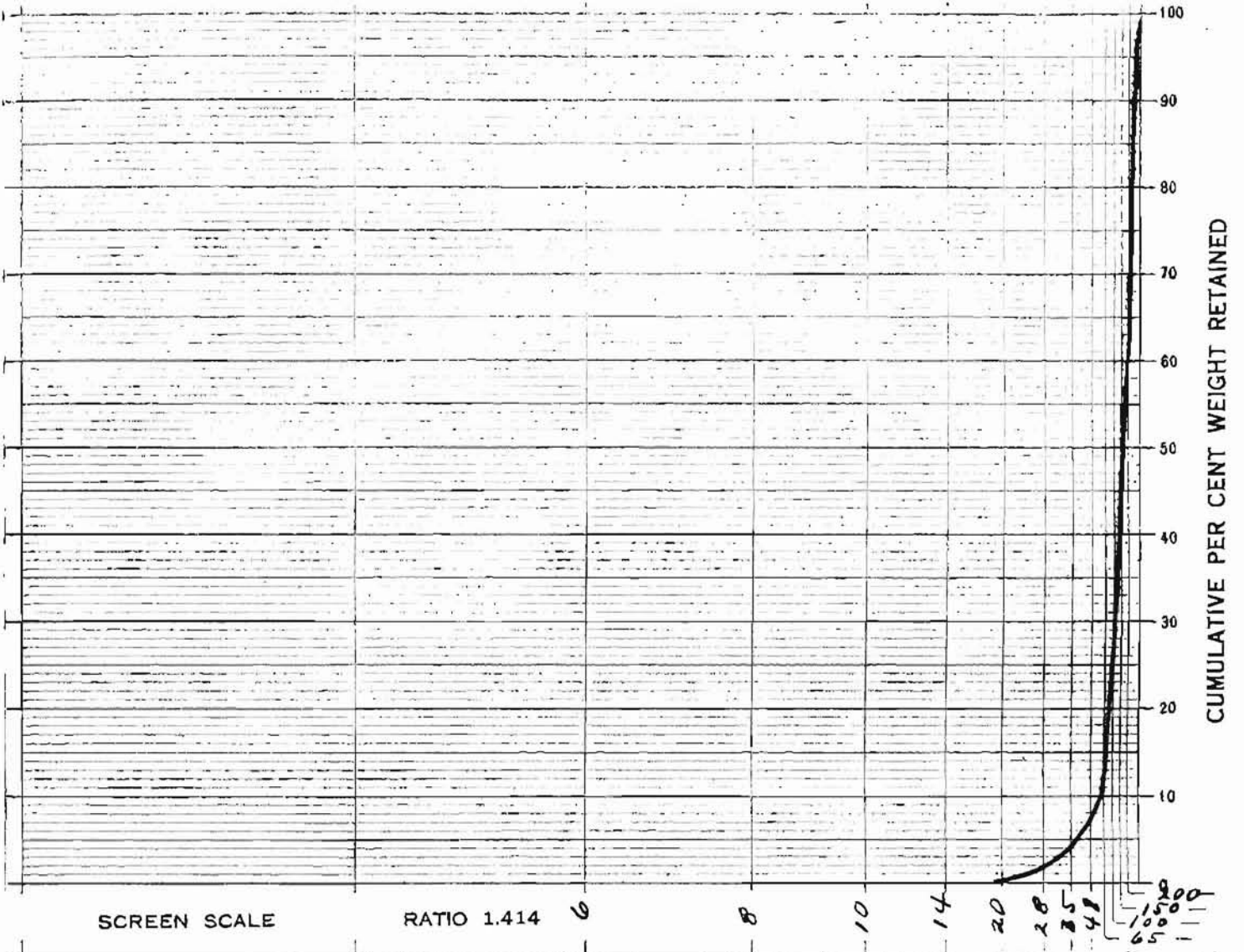
is the Screen used through also First Retaining Screen	SCREEN SCALE RATIO 1.414				Sample Weights	Per Cent	Per Cent Cumulative Weights	%
	Openings	Mesh	Diameter Wire Inches	Inches				
	1.050				26.67		.148	Gms.
	.742	18.85		.136				
	.525	13.33		.105				
	.371	9.433		.092	3.00 gm sample			
	.263	6.680	3	.070				
	.185	4.699	4	.058				
	.131	3.327	6	.036				
	.093	2.362	8	.029				
on	.065	1.651	10	.026	0.2	0.1	0.1	—
	.046	1.168	14	.025	1.6	0.5	0.6	—
	.0328	.838	20	.0172	13.4	9.5	5.1	1.8
	.0232	.589	28	.0126	55.2	11.4	23.5	1.5
	.0164	.417	35	.0122	61.6	30.5	44.0	1.4
	.0116	.295	48	.0092	61.1	20.4	64.4	1.6
	.0082	.208	65	.0073	51.5	17.1	81.5	2.0
	.0056	.147	100	.0042	40.2	13.4	94.9	3.0
	.0041	.104	150	.0028	11.7	3.9	98.8	11.4
Thru	.0029	.074	300	.0021	1.6	0.5	99.3	22.8
	.0029	.074	300	.0021	1.1	0.1	99.4	—
			Totals		299.2	99.4		

The Tyler Standard Screen Scale

Cumulative Direct Diagram of Screen Analysis on Sample of Slimes

Name _____

Date _____



No. of the Screen used through also First retaining Screen	SCREEN SCALE RATIO 1.414			Diameter Wire Inches	Sample Weights	Per Cent	Per Cent Cumulative Weights	%	Zn.	
	Inches	Milli-meters	Mesh							
	1.050	26.67		.149	<i>Gms.</i>					
	.742	18.85		.135						
	.525	13.33		.106	<i>300 gm. sample</i>					
	.371	9.423		.092						
	.263	6.680	3	.070						
	.185	4.809	4	.065						
	.131	3.327	6	.038						
	.093	2.362	8	.032						
	.065	1.651	10	.025						
<i>On</i>	.048	1.188	14	.0172	0.3	0.1	0.1	—	—	
	.0328	.833	20	.0125	0.9	0.3	0.4	—	—	
	.0233	.589	28	.0092	4.3	1.4	1.8	1.6	1.6	
	.0164	.417	35	.0065	6.9	2.3	4.7	1.6	1.8	
	.0118	.295	48	.0042	7.3	2.4	6.5	2.1	2.6	
	.0082	.208	65	.0028	10.4	3.5	10.0	2.1	3.7	
	.0058	.147	100	.0021	31.9	10.6	20.6	2.1	7.7	
	.0041	.104	150	.0015	70.9	23.6	44.2	3.3	7.7	
<i>Thru</i>	.0029	.074	200	.0011	96.7	32.2	76.4	3.3	3.3	
	.0020	.051	280	.0008	69.9	23.3	99.7			
	Totals.				299.5	99.7				

Experiment Two

For the runs which followed, it was decided to eliminate the jigs and grind the chats as they were received, in the ball mill. The question to be considered on this method was whether or not the material (the gangue being hard chert) would be too hard to grind, thus causing a consumption of too much power and too great a wear on the balls in the ball mill. For this method the chats were treated as follows: 300 lbs of the chats was weighed up and ground in the ball mill and classified, the spigot product being treated on the tables and the over-flow by flotation. A flow sheet of this operation with results is found on page 11.

These results give a table extraction of 38%. With a flotation extraction of 93%, we obtained a total extraction of 60%. From this figure, a recovery of 8.93 lbs. per 300 lbs. of chats treated was realized. This gives a total recovery of 34.2 lbs. of zinc per ton of chats treated.

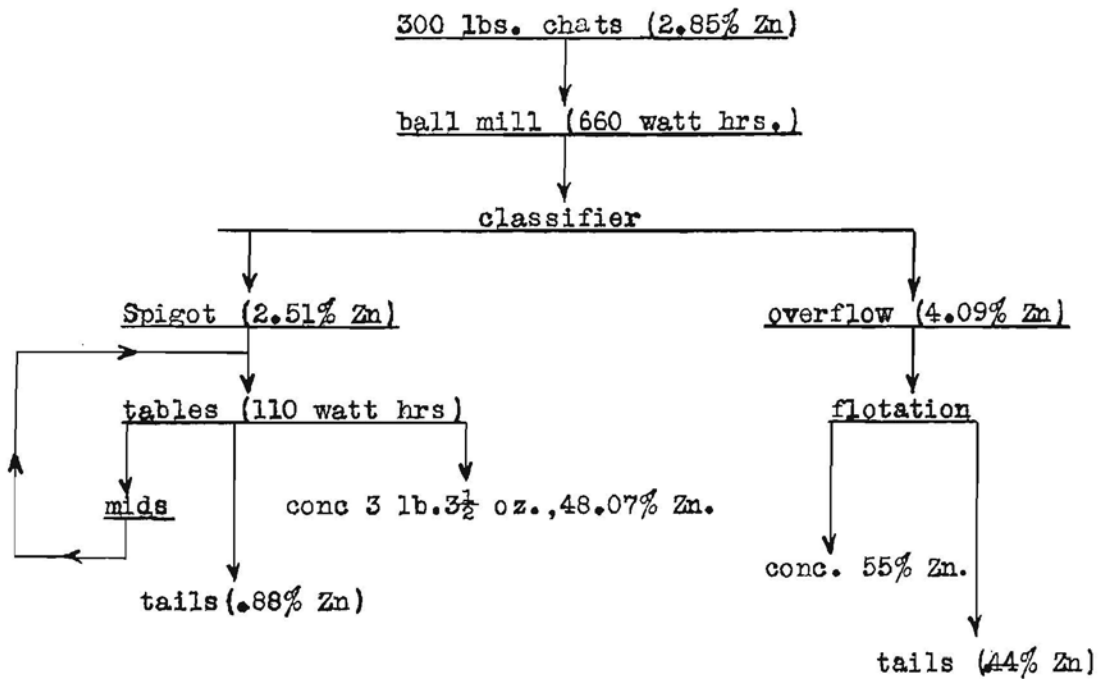
Again estimating the cost of power to be two cents per KW hour, a total cost for power amounts to .299 cents per pound of zinc recovered. The total loss of iron on the ball mill for this method was found to be about 4 lbs. per ton of chats treated.

Experiment Three

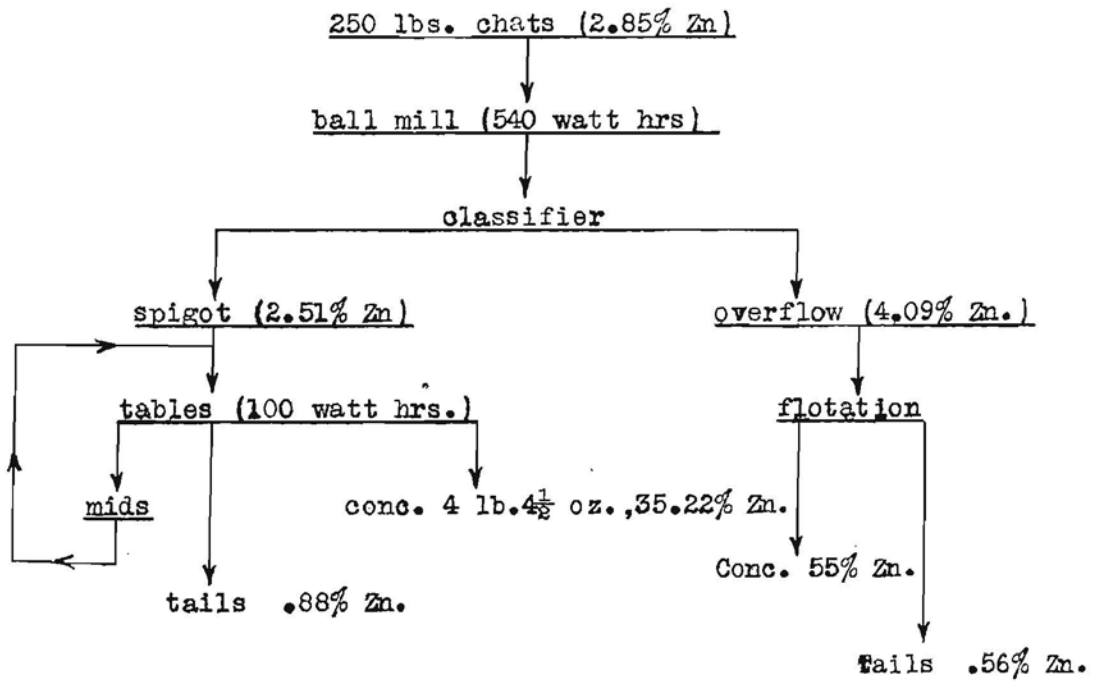
The third run was a duplication of the preceding one, except that 250 lbs. of chats was treated. A flow sheet of this operation, with results, is found on page 12. From these results, a table concentration of 69% was realized, and with a flotation

concentration of 93% a total concentration of 82% was obtained, or a total recovery of 46.74 lbs. of zinc per ton of chats treated. Using previous figures, a cost for power of .219 cents per pound of zinc recovered was calculated.

Flow Sheet Experiment No. Two.



Flow Sheet Experiment No. Three.



Experiment Four

On the last run 200 lbs. of the tailings from the rougher jigs and 100 lbs. of the tailings from the table were used. A mixture of this ratio gives a true sample of the chats from the tailings piles. An assay of this material showed it to contain 3.62% zinc.

The chats from the rougher jigs were passed through the rolls and then mixed with those from the tables. The material was then ground in the ball mill and classified. The spigot and the overflow products were treated as previously stated. A screen analysis and a flow sheet with results for this experiment will be found on pages 14 and 15 respectively.

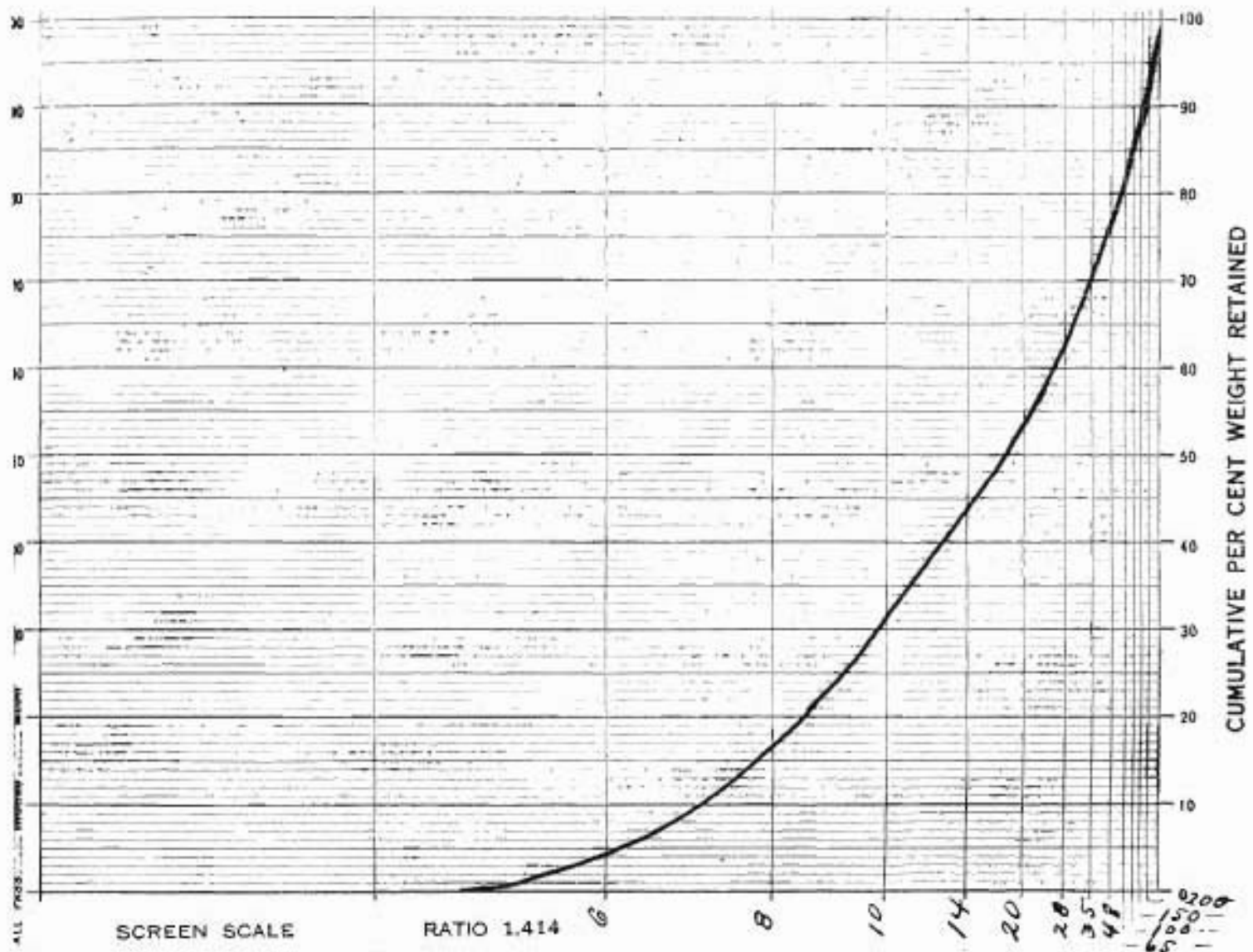
From these results a table extraction of 67% is obtained, and with a flotation extraction of 94% a total extraction of 83% results. Therefore, a recovery of 60.09 lbs. of zinc for each ton of chats treated is obtained. The cost for power in this operation amounted to .199 cents per pound of zinc recovered.

The Tyler Standard Screen Scale

Cumulative Direct Diagram of Screen Analysis on Sample of Run No. 4 from rolls

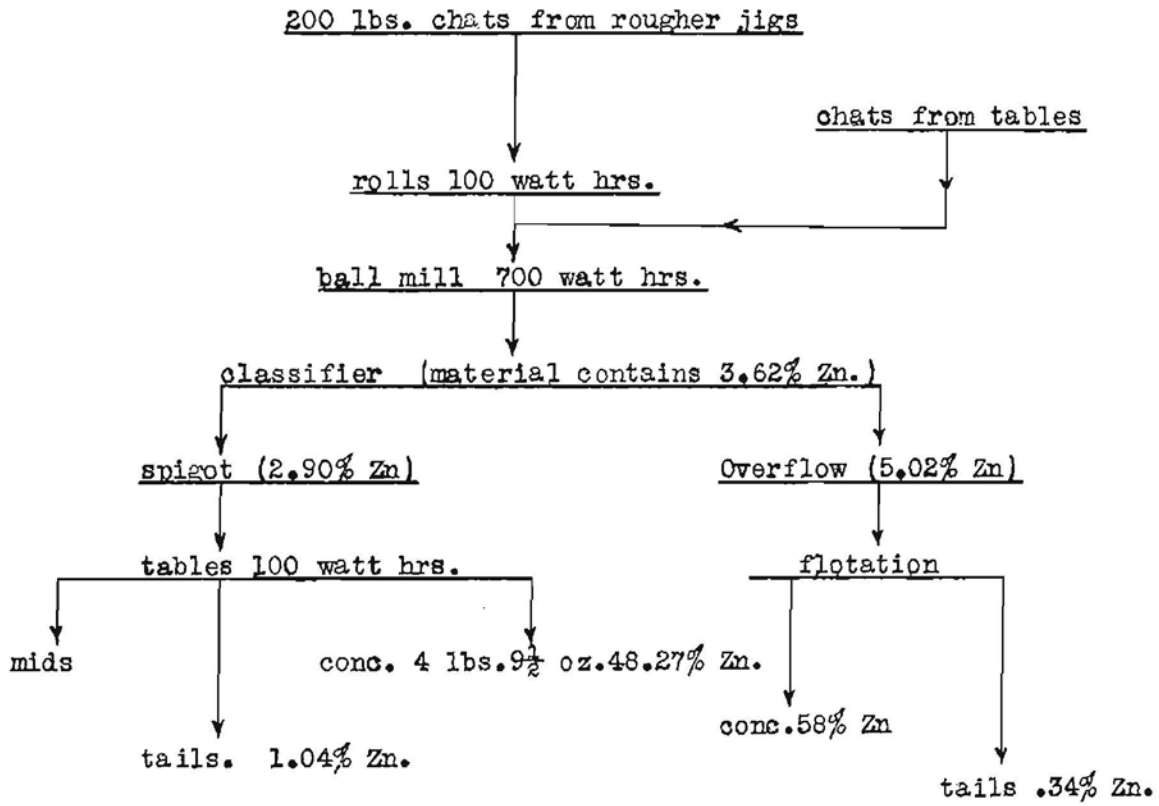
Name _____

Date _____



Inches	Millimeters	Mesh	Diameter Wire Inches	SCREEN SCALE RATIO 1.414			%
				Sample Weights	Per Cent	Per Cent Cumulative Weights	
1.060	26.67		.149	Gms.			
.742	18.86		.135				
.526	13.33		.105	300 gm sample			
.371	9.423		.082				
.263	6.690	3	.070				
.185	4.698	4	.065				
.131	3.327	6	.038	14.3	4.7	4.7	1.24
.083	2.382	8	.032	21.7	10.5	15.2	1.72
.065	1.651	10	.028	41.1	16.0	31.2	2.12
.046	1.168	14	.025	40.1	13.3	44.5	2.33
.0328	.833	30	.0172	27.7	7.9	54.4	2.17
.0232	.589	28	.0126	30.8	10.2	64.6	2.07
.0184	.417	35	.0122	21.6	7.2	71.8	2.07
.0116	.285	48	.0092	17.4	5.8	77.6	2.67
.0082	.208	65	.0072	10.4	4.8	82.4	3.68
.0058	.147	100	.0042	13.6	4.5	86.9	4.82
.0041	.104	160	.0028	12.0	4.0	90.9	10.15
.0028	.074	200	.0021	6.0	2.0	92.9	12.57
.0020	.074	200	.0021	20.1	6.7	99.6	16.16
Totals				277.8	91.6		

Flow Sheet Experiment No. Four



Machine. *Minerals Separation*.....Experimenter. *Delaloye + Kerr*.....

EXPERIMENT STATION

Missouri School of Mines and Metallurgy

FLOTATION LABORATORY.

OPERATING DATA

ORE

*Slime from classifier
Assays 4.09% Zn**10% CuSO₄ Solution Used**725 gm
Ore**Dilution
4:1**RPM Machi
1700*

Test No	OIL			REAGENTS		FROTH		REMARKS.	RESULTS						
	No.	Amt.	Kind	Kind	Amt.	Kind	Amt.		Wt. gm	Per Ct. Zn	Wt. gm	Per Ct. Zn	Wt. gm	Per Ct. Zn	Per Ct. Ext.
1		<i>Drops</i> 5	<i>A. T.</i>	<i>CuSO₄</i>	<i>25</i>	<i>Good</i>		<i>Time - 15 Min.</i>	<i>45</i>	<i>57.3</i>	<i>6</i>	<i>9.7</i>	<i>674</i>	<i>0.47</i>	<i>89</i>
2		7	"	"	40	"		" - 15 "	41	59.2	8	18.3	676	0.56	87
3		11	"	"	40	"		" - 16 "	43	58.7	17	9.6	665	0.36	96
4		6	"	"	16	"		" - 14 "	44	54.8	6	14.8	675	0.68	84
5		8	"	"	30	"		" - 15 "	53	52.3	7	10.5	665	0.27	96
6		8	"	"	50	"		" - 16 "	51	53.9	11	7.3	663	0.31	95

Missouri School of Mines and Metallurgy

FLOTATION LABORATORY.

OPERATING DATA

ORE

*Slime from classifier.
Assays 4.09% Zn**10% CuSO₄ Solution Used**725 gm
Ore**Dilution
4:1**R.P.M.
1700*

Test No	OIL			REAGENTS		FROTH		REMARKS.	RESULTS						
	No.	Amt.	Kind	Kind	Amt.	Kind	Amt.		Wt.	Per Ct.	Wt.	Per Ct.	Wt.	Per Ct.	Per Ct. Ext.
		<i>Drops</i>			<i>cc</i>				<i>gm</i>	<i>Zn</i>	<i>gm</i>	<i>Zn</i>	<i>gm</i>	<i>Zn</i>	
7		20	A.T.	CuSO ₄	60	Good		<i>Time - 20 Min</i>	47	56.8	12	12.7	666	0.34	95
8		10	"		35	"		<i>" - 20 "</i>	53	53.0	5	8.4	667	0.33	95
17 9		12	"		15	"		<i>" - 16 "</i>	55	50.6	8	9.2	662	0.26	96
10		9	"		20	"		<i>" - 20 "</i>	54	51.9	25	4.1	646	0.10	98
11		6	"		25	"		<i>" - 12 "</i>	52	55.0	7	7.1	666	0.14	97
12		7	"		30	"		<i>" - 15 "</i>	48	55.3	7	13.0	670	0.42	92

Machine. *Minerals... Separation*Experimenter. *Dalaloye + Kerr*

EXPERIMENT STATION

Missouri School of Mines and Metallurgy

FLOTATION LABORATORY.

OPERATING DATA

ORE
Slime from classifier
*Assays 4.09% Zn**10% Solution CuSO₄ Used*725 gm
Ore Dilution
 4:1 R.P.M
 1700

Test No	OIL			REAGENTS		FROTH		REMARKS.	RESULTS						
	No.	Amt.	Kind	Kind	Amt.	Kind	Amt.		Wt. gm	Per Ct. Zn	Wt. gm	Per Ct. Zn	Wt. gm	Per Ct. Zn	Per Ct. Ext.
13		<i>Drops</i> 7	<i>A.T.</i>	<i>CuSO₄</i>	<i>40</i>	<i>Good</i>		<i>Time — 15 Min.</i>	<i>47</i>	<i>55.8</i>	<i>9</i>	<i>10.2</i>	<i>669</i>	<i>0.34</i>	<i>95</i>
14		7	"		50	"		" — 14 "	48	53.1	17	6.9	660	0.43	90
15		9	"		60	"		" — 15 "	50	52.9	12	8.6	663	0.40	91

Machine *Minerals Separation*Experimenter *Delaloye + Kerr*

EXPERIMENT STATION

Missouri School of Mines and Metallurgy

FLOTATION LABORATORY.

OPERATING DATA

ORE

*Slime from classifier
Assays 5.02% Zn**10% CuSO₄ Solution Used**725 gm
Ore**Dilution
4:1**R.P.M.
1700*

TestNo	OIL			REAGENTS		FROTH		REMARKS.	RESULTS						
	No.	Amt.	Kind	Kind	Amt.	Kind	Amt.		Wt.	Per Ct.	Wt.	Per Ct	Wt.	Per Ct	Per Ct. Ext.
		<i>Drops</i>			<i>cc</i>				<i>gm</i>	<i>Zn</i>	<i>gm</i>	<i>Zn</i>	<i>gm</i>	<i>Zn</i>	
16		7	A.T.	CuSO ₄	30	Good		<i>Time - 15 Min.</i>	52	53.5	14	22.9	659	0.54	88
17		6	"	"	20	"		<i>" - 15 "</i>	50	57.1	19	24.7	656	0.09	97
-19- 18		6	"	"	10	"		<i>" - 15 "</i>	48	58.7	23	36.1	654	0.14	97
19		7	"	"	10	"		<i>" - 15 "</i>	38	60.0	17	39.3	670	0.68	87
20		6	"	"	10	"		<i>" - 15 "</i>	52	59.7	12	32.9	661	0.07	98
21		6	"	"	10	"		<i>" - 15 "</i>	44	58.4	12	39.6	669	0.54	90

Conclusions

In drawing conclusions, the last experiment will have to be referred to, as it represents the treatment of a true sample of the chats that are discarded in the district. The results obtained from the tables are not as good as those obtained from flotation; but, referring to the work described on page one, the critical crushing point of this material is through 65-mesh. An average of these experiments shows that about 40% of the material classified is spigot product and must be tabled; the remaining 60% must be floated. The results of the experiments on the flotation of these chats are to be found on pages 16 to 19 inclusive.

From these results one can see that a good extraction can be obtained by flotation. Experimental work on the flotation of these chats showed that the best oil to use was the A. T. mixture, manufactured by the Newport Chemical Company, Passaic, N. J. The only other reagent that was used was copper sulfate, the amount of which was varied in the different charges. This amount was varied from 6 cc to 60 cc of a 10% solution, and it was found that the best results were obtained when 10 cc were used. This would amount to 1 pound of copper sulfate per ton of chats treated. The amount of oil was also varied, but it was found that 6 drops per charge of 700 gms. was sufficient, or 6 lbs. per ton.